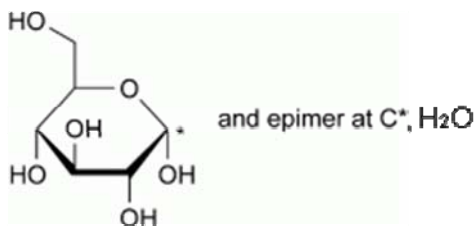


E-56 GLUCOSE MONOHYDRATE

Stage 4 rev.



M_r 198.2

$C_6H_{12}O_6 \cdot H_2O$

DEFINITION

Glucose monohydrate is the monohydrate of (+)-D-glucopyranose and is derived from starch.

Content: 97.5 per cent to 102.0 per cent (anhydrous substance), determined by the LC described in the assay.

IDENTIFICATION

A. Infrared absorption spectrophotometry.

Record the infrared absorption spectrum of glucose monohydrate and compare with the Reference Spectrum or the spectrum obtained with the Reference Standard: the transmission minima correspond in position and relative size.

B. Examine the chromatograms obtained in the assay.

The principal peak in the chromatogram obtained with the test solution is similar in retention time and size to the principal peak in the chromatogram obtained with reference solution (a).

C. Water (see tests).

TESTS

Appearance of solution. The test solution is clear (its clarity is the same as that of water or its opalescence is not more pronounced than that of reference suspension I) and not more intensely coloured than the reference solution.

Test solution: Dissolve 10.0 g in 15 mL of water.

Primary solutions:

- *Ferric chloride primary solution:* a 45.0 g/l solution of ferric chloride ($FeCl_3 \cdot 6H_2O$).
- *Cobalt chloride primary solution:* a 59.5 g/l solution of cobalt chloride ($CoCl_2 \cdot 6H_2O$).
- *Copper sulfate primary solution:* a 62.4 g/l solution of copper sulfate ($CuSO_4 \cdot 5H_2O$).

45

46 *Reference solution:* to 2.5 mL of cobalt chloride primary solution, 6.0 mL of ferric
47 chloride primary solution and 1.0 mL of copper sulfate primary solution, add
48 hydrochloric acid (10 g/l HCl) to make 1000.0 mL.

49

50 **Conductivity:** maximum $20 \mu\text{S}\cdot\text{cm}^{-1}$ at 25°C.

51 Dissolve 20.0 g in *carbon dioxide-free water* prepared from *distilled water* and dilute
52 to 100.0 mL with the same solvent. Measure the conductivity of the solution while
53 gently stirring with a magnetic stirrer.

54

55 **Related substances.** Liquid chromatography.

56

57 *Test solution.* Dissolve 0.330 g of the substance to be examined in *water*, and dilute to
58 10.0 mL with the same solvent.

59

60 *Reference solution (a).* Dissolve 0.330 g of *glucose monohydrate CRS* in *water* and
61 dilute to 10.0 mL with the same solvent.

62

63 *Reference solution (b).* Dilute 1.0 mL of the test solution to 250.0 mL with *water*.

64

65 *Reference solution (c).* Dilute 25.0 mL of reference solution (b) to 200.0 mL with
66 *water*.

67

68 *Reference solution (d).* Dissolve 5 mg of *maltose* (impurity A), 5 mg of *maltotriose*
69 (impurity C) and 5 mg of *fructose* (impurity D) in *water* and dilute to 50.0 mL with
70 *water*.

71

72 *Column:*

73 - size: $l = 0.3 \text{ m}$, $\varnothing = 7.8 \text{ mm}$;

74 - stationary phase: strong cation-exchange resin (calcium form)¹ (9 μm);

75 - temperature: 85 +/- 1°C.

76

77 *Mobile phase:* degassed water.

78

79 *Flow rate:* 0.3 mL/min.

80

81 *Detection:* refractometer maintained at a constant temperature (40°C for example).

82

83 *Injection:* 20 μl of the test solution and reference solutions (b), (c) and (d).

84

85 *Run time:* 1.5 times the retention time of glucose.

86

87 *Relative retention* with reference to glucose (retention time = about 21 min):
88 impurity C = about 0.7; impurities A and B = about 0.8; impurity D = about 1.3.

89

90 *System suitability* : reference solution (d) :

91 - *resolution* : minimum 1.3 between the peaks due to impurities C and A.

92

¹ Aminex HPX-87C from Biorad is suitable.

93 *Limits:*

- 94 - sum of impurities A and B: not more than the area of the principal peak in the
 95 chromatogram obtained with reference solution (b) (0.4 per cent),
 96 - impurity C: not more than 0.5 times the area of the principal peak in the
 97 chromatogram obtained with reference solution (b) (0.2 per cent),
 98 - impurity D: not more than 3 times the area of the principal peak in the
 99 chromatogram obtained with reference solution (c) (0.15 per cent),
 100 - unspecified impurities for each impurity, not more than twice the area of the
 101 principal peak in the chromatogram obtained with reference solution (c) (0.10 per
 102 cent),
 103 - total: not more than 1.25 times the area of the principal peak in the chromatogram
 104 obtained with reference solution (b) (0.5 per cent),
 105 - disregard limit: area of the principal peak in the chromatogram obtained with
 106 reference solution (c) (0.05 per cent).

107

108 **Dextrin.** Reflux 1 g of the substance to be examined finely powdered with 20 mL of
 109 ethanol (96 per cent): it dissolves completely.

110

111 **Soluble starch, sulphite:** maximum 15 ppm.

112 Dissolve 10.0 g in 15 mL of *water* using a bath of boiling water. Allow to cool and
 113 add 50 μ l of 0.1 N iodine: the solution is yellow.

114

115 **Water:** 7.5 per cent to 9.5 per cent, determined on 0.25 g by the semi-micro
 116 determination of water.

117

118 ASSAY

119

120 Liquid chromatography as described in the test for related substances with the
 121 following modification.

122

123 *Injection:* test solution and reference solution (a).

124

125 Calculate the percentage content of $C_6H_{12}O_6$ from the areas of the peaks and the
 126 assigned content of *glucose monohydrate CRS*.

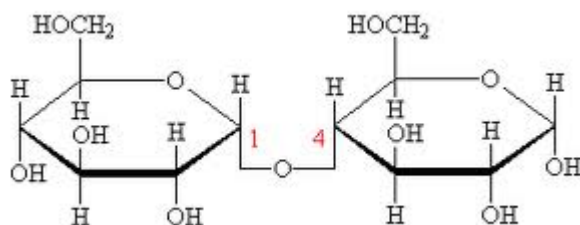
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128 IMPURITIES

129

130 A. Maltose

131



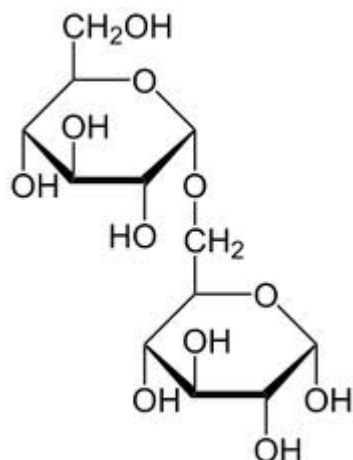
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135 B. Isomaltose

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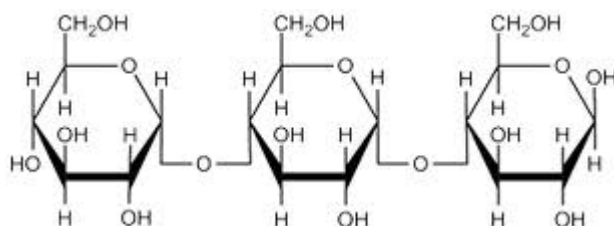
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140 C. Maltotriose

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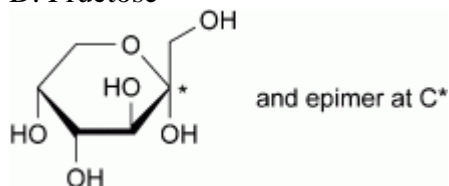


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145 D. Fructose



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149

REAGENTS

150

151 **Hydrazine sulfate solution.** Dissolve 1.0 g of hydrazine sulfate in water and dilute to
152 100.0 mL with the same solvent. Allow to stand for 4-6 h.

153

154 **Hexamethylenetetramine solution.** In a 100 mL ground-glass-stoppered flask,
155 dissolve 2.5 g of hexamethylenetetramine in 25.0 mL of water.

156

157 **Primary opalescent suspension (formazin suspension).** To the
158 hexamethylenetetramine solution in the flask add 25.0 mL of the hydrazine sulfate
159 solution. Mix and allow to stand for 24 h. This suspension is stable for 2 months,
160 provided it is stored in a glass container free from surface defects. The suspension
161 must not adhere to the glass and must be well mixed before use.

162

163 **Standard of opalescence.** Dilute 15.0 mL of the primary opalescent suspension to
164 1000.0 mL with water. This suspension is freshly prepared and may be stored for up
165 to 24 h.

166

167 **Reference suspension I.** To 5.0 mL of standard of opalescence add 95.0 mL of water.
168 Mix and shake before use.

169

170 **Cation exchange resin (calcium form), strong.**

171 A resin in calcium form with sulfonic acid groups attached to a polymer lattice
172 consisting of polystyrene cross-linked with 8 per cent of divinylbenzene. The particle
173 size is specified after the name of the reagent in the tests where it is used.

174

175 **Fructose.** $C_6H_{12}O_6$. (M_r 180.2). [57-48-7].

176

177 **Maltose monohydrate.** $C_{12}H_{22}O_{11}$, H_2O . (M_r 360.3). [6363-53-7].

178

179 **Maltotriose.** $C_{18}H_{32}O_{16}$. (M_r 504.4). [1109-28-0].

180

181