

1 Phenobarbital Tablets

2 フェノバルビタール錠

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4 Phenobarbital Tablets contain not less than 95.0%
5 and not more than 105.0% of the labeled amount of
6 phenobarbital ($C_{12}H_{12}N_2O_3$; 232.24).

7 **Method of preparation** Prepare as directed under Tab-
8 lets, with Phenobarbital.

9 **Identification** To a quantity of powdered Phenobarbital
10 Tablets, equivalent to 20 mg of Phenobarbital, add 20 mL
11 of boric acid-potassium chloride-sodium hydroxide buffer
12 solution (pH 9.6), shake, and centrifuge. To 1 mL of the
13 supernatant liquid add boric acid-potassium chloride-so-
14 dium hydroxide buffer solution (pH 9.6) to make 100 mL.
15 Determine the absorption spectrum of this solution as di-
16 rected under Ultraviolet-visible Spectrophotometry <2.24>:
17 it exhibits a maximum between 238 nm and 242 nm.

18 **Uniformity of dosage units** <6.02> Perform the Mass
19 variation test, or the Content uniformity test according to
20 the following method: it meets the requirement.

21 To 1 tablet of Phenobarbital Tablets add exactly V mL of
22 a mixture of water and acetonitrile (1:1) so that each mL
23 contains about 1 mg of phenobarbital ($C_{12}H_{12}N_2O_3$), soni-
24 cate to disintegrate, shake for 10 minutes, and centrifuge.
25 Then, proceed as directed in the Assay.

$$26 \quad \text{Amount (mg) of phenobarbital (C}_{12}\text{H}_{12}\text{N}_2\text{O}_3\text{)} \\ 27 \quad = M_S \times A_T / A_S \times V / 30$$

28 M_S : Amount (mg) of phenobarbital for assay taken

29 **Dissolution** <6.10> When the test is performed at 50 rev-
30 olutions per minute according to the Paddle method, using
31 900 mL of water as the dissolution medium, the dissolution
32 rate in 30 minutes of Phenobarbital Tablets is not less than
33 75%.

34 Start the test with 1 tablet of Phenobarbital Tablets, with-
35 draw not less than 20 mL of the medium at the specified
36 minute after starting the test, and filter through a membrane
37 filter with a pore size not exceeding 0.45 μm . Discard the
38 first 10 mL of the filtrate, pipet V mL of the subsequent
39 filtrate, and add water to make exactly V' mL so that each
40 mL contains about 33 μg of phenobarbital ($C_{12}H_{12}N_2O_3$).
41 Pipet 5 mL of this solution, add exactly 10 mL of boric
42 acid-potassium chloride-sodium hydroxide buffer solution
43 (pH 9.6), and use this solution as the sample solution. Sep-
44 arately, weigh accurately about 17 mg of phenobarbital for
45 assay, previously dried at 105°C for 2 hours, and dissolve
46 in water to make exactly 100 mL. Pipet 5 mL of this solu-
47 tion, and add water to make exactly 25 mL. Pipet 5 mL of

48 this solution, add exactly 10 mL of boric acid-potassium
49 chloride-sodium hydroxide buffer solution (pH 9.6), and
50 use this solution as the standard solution. Determine the ab-
51 sorbances, A_T and A_S , at 240 nm of the sample solution and
52 standard solution as directed under Ultraviolet-visible
53 Spectrophotometry <2.24>, using a mixture of boric acid-
54 potassium chloride-sodium hydroxide buffer solution (pH
55 9.6) and water (2:1) as the blank.

56 Dissolution rate (%) with respect to the labeled amount of
57 phenobarbital ($C_{12}H_{12}N_2O_3$)

$$58 \quad = M_S \times A_T / A_S \times V' / V \times 1 / C \times 180$$

59 M_S : Amount (mg) of phenobarbital for assay taken

60 C : Labeled amount (mg) of phenobarbital ($C_{12}H_{12}N_2O_3$)
61 in 1 tablet

62 **Assay** Weigh accurately the mass of not less than 20 tab-
63 lets of Phenobarbital Tablets, and powder. Weigh accu-
64 rately a portion of the powder, equivalent to about 30 mg
65 of phenobarbital ($C_{12}H_{12}N_2O_3$), add exactly 30 mL a mix-
66 ture of water and acetonitrile (1:1), shake for 10 minutes,
67 and centrifuge. Pipet 1 mL of the supernatant liquid, add a
68 mixture of water and acetonitrile (1:1) to make exactly 20
69 mL, and use this solution as the sample solution. Separately,
70 weigh accurately about 30 mg of phenobarbital for assay,
71 previously dried at 105°C for 2 hours, and dissolve in a
72 mixture of water and acetonitrile (1:1) to make exactly 30
73 mL. Pipet 1 mL of this solution, add a mixture of water and
74 acetonitrile (1:1) to make exactly 20 mL, and use this solu-
75 tion as the standard solution. Perform the test with exactly
76 10 μL each of the sample solution and standard solution as
77 directed under Liquid Chromatography <2.01> according to
78 the following conditions, and determine the peak areas, A_T
79 and A_S , of phenobarbital in each solution.

$$80 \quad \text{Amount (mg) of phenobarbital (C}_{12}\text{H}_{12}\text{N}_2\text{O}_3\text{)} \\ 81 \quad = M_S \times A_T / A_S$$

82 M_S : Amount (mg) of phenobarbital for assay taken

83 *Operating conditions*—

84 **Detector:** An ultraviolet absorption photometer
85 (wavelength: 210 nm).

86 **Column:** A stainless steel column 4.6 mm in inside
87 diameter and 15 cm in length, packed with
88 octadecylsilanized silica gel for liquid chromatography (5
89 μm in particle diameter).

90 **Column temperature:** A constant temperature of about
91 45°C.

92 **Mobile phase:** A mixture of water and acetonitrile for
93 liquid chromatography (11:9).

94 **Flow rate:** Adjust so that the retention time of
95 phenobarbital is about 3 minutes.

96 *System suitability*—

97 System performance: When the procedure is run with 10
98 μL of the standard solution under the above operating
99 conditions, the number of theoretical plates and the
100 symmetry factor of the peak of phenobarbital are not less
101 than 3000 and not more than 1.5, respectively.

102 System repeatability: When the test is repeated 6 times
103 with 10 μL of the standard solution under the above
104 operating conditions, the relative standard deviation of the
105 peak area of phenobarbital is not more than 1.0%.

106 **Containers and storage** Containers—Well-closed con-
107 tainers.

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