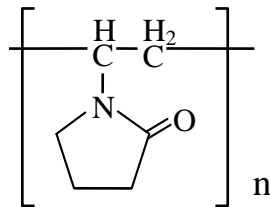


Povidone  
(Rev. 1, Stage 4)

Polyvidone  
Polyvinylpyrrolidone



$(C_6H_9NO)_n$  [9003-39-8]

Poly [(2-oxo-1-pyrrolidinyl) ethylene]

Povidone is a chain polymer of 1-vinyl-2-pyrrolidone. It contains not less than 11.5% and not more than 12.8% of nitrogen (N: 14.01), calculated on the anhydrous basis.

It has the nominal K-value of not less than 10 and not more than 120.

The nominal K-value is shown on the label.

**Identification:**

(1) To 0.5 g of Povidone, add 10 ml of water, and shake. The substance dissolves.

**pH** Dissolve 1.0 g of Povidone in 20 mL of water: the pH of this solution is between 3.0 and 5.0 for Povidone having the nominal K-value of 30 or less, and between 4.0 and 7.0 for Povidone having the nominal K-value exceeding 30.

**Purity**

(1) Aldehydes – Weigh accurately about 1.0 g of Povidone, and dissolve in 0.05 mol/L pyrophosphate buffer solution, pH 9.0 to make exactly 100 mL. Stopper tightly, warm at 60°C for 60 minutes, allow to cool to room temperature, and use this solution as the sample solution. Separately, prepare the standard solution as follows: dissolve 0.140 g of acetaldehyde ammonia trimer trihydrate in water to make 200.0 mL, dilute 1.0 mL of the solution to 100.0 mL with 0.05 mol/L pyrophosphate buffer solution, pH 9.0.

Measure exactly 0.5 mL each of the sample solution, the standard solution and water (for blank test), transfer to separate cells with a path length of 1 cm, add 2.5 mL of 0.05 mol/L pyrophosphate buffer solution, pH 9.0, and 0.2 mL of  $\beta$ -nicotinamide adenine dinucleotide TS to each of these cells, mix and stopper

41 tightly. Allow to stand for 2 to 3 minutes at  $22 \pm 2^\circ\text{C}$ , and perform the test with  
 42 these solutions as directed under the Spectrophotometry using water as the control  
 43 solution. Determine the absorbances,  $A_{t1}$ ,  $A_{s1}$  and  $A_{b1}$  of the subsequent solutions  
 44 of the sample solution, the standard solution and water (blank) at 340 nm. Then,  
 45 add 0.05 mL of aldehyde dehydrogenase TS to each of the cells, stir and stopper  
 46 tightly. Allow to stand at  $22 \pm 2^\circ\text{C}$  for 5 minutes. Determine the absorbances,  $A_{t2}$ ,  
 47  $A_{s2}$  and  $A_{b2}$  of these solutions in the same manner as above: the content of  
 48 aldehydes is not more than 500 ppm (as acetaldehyde).

49  
 50 Content (ppm) of aldehydes as acetaldehyde

$$51 \quad = \frac{(A_{t2} - A_{t1}) - (A_{b2} - A_{b1})}{(A_{s2} - A_{s1}) - (A_{b2} - A_{b1})} \times \frac{W'}{W} \times 100000$$

52  
 53 W: Weighed amount (g) of Povidone, calculated on the anhydrous basis.

54 W': Concentration (mg/mL) of acetaldehyde in the standard solution,  
 55 calculated from the weight of the acetaldehyde ammonia trimer  
 56 trihydrate with the factor 0.72.

57  
 58 (2) 1-Vinyl-2-pyrrolidone—Weigh accurately about 0.25 g of Povidone, dissolve in a  
 59 mixture of acetonitrile and water [10 : 90 (v : v)] to make exactly 10 mL, and use  
 60 this solution as the sample solution. Separately, dissolve 0.050 g of  
 61 1-vinyl-2-pyrrolidone in a mixture of acetonitrile and water [10 : 90 (v : v)] to  
 62 make exactly 100mL. Pipet 1 mL of this solution and add a mixture of acetonitrile  
 63 and water [10 : 90 (v : v)] to make exactly 100 mL. Pipet 5 mL of this solution and  
 64 add a mixture of acetonitrile and water [10 : 90 (v : v)] to make exactly 100 mL,  
 65 and use this solution as the standard solution. Perform the test with exactly 20  $\mu\text{L}$   
 66 each of the sample solution and the standard solution as directed under the Liquid  
 67 Chromatography according to the following conditions, and determine the peak  
 68 areas,  $A_T$  and  $A_S$ , of 1-vinyl-2-pyrrolidone in each solution: the content of  
 69 1-vinyl-2- pyrrolidone is not more than 10 ppm.

$$70 \quad \text{Content (ppm) of 1-vinyl-2-pyrrolidone} = \frac{A_T}{A_S} \times \frac{2.5}{W}$$

71  
 72  
 73 W: Weighed amount (g) of Povidone, calculated on the anhydrous basis.

74 *Operating conditions –*

75 Detector: An ultraviolet spectrophotometer (detection wavelength: 235 nm)

76 Column: Stainless steel column 4.0 mm in inside diameter and about 10 mm in  
 77 length, and 4.6 mm in inside diameter and about 150 mm in length, packed with

78 octadecylsilanized silica gel for liquid chromatography (5  $\mu\text{m}$  in particle  
79 diameter), and use them as a guard column and a separation column,  
80 respectively.

81

82 Column temperature: A constant temperature of about 40°C.

83 Mobile phase: Acetonitrile : water [10 : 90 (v : v)] .

84 Flow rate: 1.0 mL/min

85 Selection of column: Dissolve 0.01 g of 1-vinyl-2-pyrrolidone and 0.5 g of vinyl  
86 acetate in 100 mL of methanol. To 1 mL of this solution add a mixture of  
87 acetonitrile and water [10 : 90 (v : v)] to make 100 mL. Proceed with 50  $\mu\text{L}$  of  
88 this solution according to the above operating conditions. Use a column giving  
89 elution of 1-vinyl-2-pyrrolidone and vinyl acetate in this order with the  
90 resolution between their peaks being not less than 2.0.

91 System reproducibility: When the test is repeated six times with the standard  
92 solution under the above operating conditions, the relative standard deviation of  
93 obtained peak areas of 1-vinyl-2-pyrrolidone is not more than 2.0%.

94

95 (3) Peroxides – Weigh exactly an amount of Povidone, equivalent to 4.0 g calculated  
96 on the anhydrous basis, dissolve in water to make exactly 100 mL, and use this  
97 solution as the sample solution. To 25 mL of the sample solution add 2 mL of  
98 titanium (III) chloride-sulfuric acid TS, and mix. Allow to stand for 30 minutes,  
99 and perform the test with this solution as directed under the Spectrophotometry,  
100 using a solution prepared by adding 2 mL of 13% sulfuric acid to 25 mL of the  
101 sample solution as a blank: the absorbance of the subsequent solution of the  
102 sample solution at 405 nm is not more than 0.35 (not more than 400 ppm, as  
103 hydrogen peroxide).

104

105 (4) Hydrazine – Weigh exactly an amount of Povidone, equivalent to 2.5 g calculated  
106 on the anhydrous basis, transfer to a 50-mL centrifuge tube, add 25 mL of water,  
107 and stir to dissolve. Add 500  $\mu\text{L}$  of a solution of salicylaldehyde in methanol (1 in  
108 20), stir and warm at 60°C for 15 minutes in a water bath. Allow to cool, add 2.0  
109 mL of toluene, stopper tightly, shake vigorously for 2 minutes, centrifuge, and use  
110 the upper layer of the mixture as the sample solution. Separately, dissolve 0.09 g  
111 of salicylaldazine in toluene to make exactly 100 mL. Pipet 1 mL of this solution,  
112 add toluene to make exactly 100 mL, and use this solution as the standard solution.  
113 Perform the test with these solutions as directed under the Thin-layer  
114 Chromatography. Spot 10  $\mu\text{L}$  each of the sample solution and the standard  
115 solution on a plate coated with a 0.25 mm layer of dimethylsilanized silica gel  
116 with fluorescent indicator for thin-layer chromatography. Develop the plate with a  
117 mixture of methanol and water (2 : 1) to a distance of about three-fourths of the

118 length of the plate, and air-dry the plate. Examine under ultraviolet light (main  
119 wavelength 365 nm) : the *R<sub>f</sub>* value of the fluorescent spot from the standard  
120 solution is about 0.3, and the fluorescence of the spot from the sample solution  
121 corresponding to the spot from the standard solution is not more intense than that  
122 of the spot from the standard solution (not more than 1 ppm).

123 (5) Formic acid – Weigh accurately about 2.0 g of Povidone, dissolve in water to  
124 make exactly 100 mL, and use this solution as the sample stock solution.  
125 Transfer a suspension of strongly acidic ion exchange resin (H<sup>+</sup> type) for column  
126 chromatography in water to a column of about 0.8 cm in inside diameter to give a  
127 packing depth of about 20 mm in length, and keep the strongly acidic ion  
128 exchange resin layer constantly immersed in water. Pour 5 mL of water, and adjust  
129 the flow rate about 1 mL/min. When the level of the water comes down to near the  
130 top of the strongly acidic ion exchange resin layer, put the sample stock solution  
131 into the column. After dropping 2 mL of the solution, collect 1.5 mL of the  
132 solution, and use this solution as the sample solution. Separately, dissolve 0.100 g  
133 of formic acid in water to make exactly 100 mL. Pipet 1 mL of this solution and  
134 add water to make exactly 100 mL, and use this solution as the standard solution.  
135 Perform the test with exactly 50 µL each of the sample solution and the standard  
136 solution as directed under the Liquid Chromatography according to the following  
137 conditions, and determine the peak areas, *A<sub>T</sub>* and *A<sub>S</sub>*, of formic acid in each  
138 solution: the content of formic acid is not more than 0.5%  
139

$$140 \quad \text{Content(\%)} \text{ of formic acid} = \frac{A_T}{A_S} \times \frac{1.0}{W \times 10}$$

141

142 W: Weighed amount (g) of Povidone, calculated on the anhydrous basis.

143 *Operating conditions*

144 Detector: An ultraviolet spectrophotometer (detection wavelength: 210 nm)

145 Column: Stainless steel columns 7.8 mm in inside diameter and 300 mm in  
146 length, packed with strongly acidic ion exchange resin for liquid  
147 chromatography ( about 10 µm in particle diameter).

148 Column temperature: A constant temperature of about 35°C

149 Mobile phase: Diluted perchloric acid (1 in 700).

150 Flow rate: 1.0 mL/min.

151 System reproducibility: When the test is repeated six times with the standard  
152 solution under the above operating conditions, the relative standard deviation of  
153 obtained peak areas of formic acid is not more than 2.0%.

154

155 (6) 2-Pyrrolidone – Weigh accurately about 0.5 g of Povidone, dissolve in a mixture

156 of water and methanol [19:1 (v:v)] to make exactly 100 mL, and use this solution  
 157 as the sample solution. Separately, dissolve 0.150 g of 2-pyrrolidone in a mixture  
 158 of water and methanol [19:1 (v:v)] to make exactly 100 mL. Pipet 2 mL of this  
 159 solution, add a mixture of water and methanol [19:1 (v:v)] to make exactly 100  
 160 mL, and use this solution as the standard solution. Perform the test with exactly 50  
 161  $\mu$ L each of the sample solution and the standard solution as directed under the  
 162 Liquid Chromatography according to the following conditions, and determine the  
 163 peak areas,  $A_T$  and  $A_S$ , of 2-pyrrolidone in each solution: the content of  
 164 2-pyrrolidone is not more than 3.0%.

165

$$166 \quad \text{Content (\% of 2-pyrrolidone)} = \frac{A_T}{A_S} \times \frac{0.3}{W}$$

167

168 W: Weighed amount (g) of Povidone, calculated on the anhydrous basis.

169 *Operating conditions –*

170 Detector: An ultraviolet spectrophotometer (detection wavelength: 205 nm)

171 Column: Stainless steel column 4.0 mm in inside diameter and about 10 mm in  
 172 length, and 4.6 mm in inside diameter and about 150 mm in length, packed  
 173 with octadecylsilanized silica gel for liquid chromatography (5  $\mu$ m in particle  
 174 diameter), and use them as a guard column and a separation column,  
 175 respectively.

176 Column temperature: A constant temperature of about 40°C.

177 Mobile phase: Water : Methanol [19 : 1 (v : v)]

178 Flow rate: 0.8 mL/min

179 System reproducibility: When the test is repeated six times with the standard  
 180 solution under the above operating conditions, the relative standard deviation  
 181 of obtained peak areas of 2-pyrrolidone is not more than 2.0%.

182 **Water** Not more than 5.0 % (0.5 g) by Karl Fischer method.

183 **Residue on ignition** Not more than 0.1 % (1 g).

184 **K-value** Weigh accurately an amount of Povidone, calculated on the anhydrous  
 185 basis, specified in the following table, dissolve in water to make exactly 100 mL,  
 186 allow to stand for 60 minutes, and use this solution as the sample solution. Perform  
 187 the test with the sample solution and with water at 25°C as directed in Method 1 under  
 188 the Viscosity Determination, and calculate the K-value by the following formula.  
 189 The K-value of Povidone having a nominal K-value of 15 or less is not less than  
 190 85.0 % and not more than 115.0 % of the nominal K-value, and the K-value of  
 191 Povidone having a nominal K-value exceeding 15 is not less than 90.0 % and not more

192 than 108.0 % of the nominal K-value.

193

$$194 \quad K = \frac{1.5 \log v_{\text{rel.}} - 1}{0.15 + 0.003c} + \frac{\sqrt{300c \log v_{\text{rel.}} + (c + 1.5c \log v_{\text{rel.}})^2}}{0.15c + 0.003c^2}$$

195

196  $c$ : Mass (g) of Povidone in 100 mL of the solution, calculated on the  
197 anhydrous basis.

198  $v_{\text{rel.}}$ : Kinematic viscosity of the sample solution relative to that of water.

199

	Nominal K-value	g
200	18 or less	5.00
201	more than 18 and not more than 95	1.00
202	more than 95	0.10

203

204  
205 **Assay** Weigh accurately about 0.1 g of Povidone, and place in a Kjeldahl flask.  
206 Add 5 g of a powdered mixture of 33 g of potassium sulfate, 1 g of cupric sulfate and  
207 1 g of titanium dioxide, and wash down any adhering sample from the neck of the  
208 flask with a small amount of water. Add 7 mL of sulfuric acid allowing to flow down  
209 the inside wall of the flask. Heat the flask gradually until the solution has a clear,  
210 yellow-green color, and the inside wall of the flask is free from a carbonized material,  
211 and then heat for further 45 minutes. After cooling, add cautiously 20 mL of water, and  
212 connect the flask to the distillation apparatus previously washed by passing steam  
213 through it. To the absorption flask add 30 mL of a solution of boric acid (1 in 25), 3  
214 drops of bromocresol green-methyl red TS and sufficient water to immerse the lower  
215 end of the condenser tube. Add 30 mL of a solution of sodium hydroxide (2 in 5)  
216 through the funnel, rinse cautiously the funnel with 10 mL of water, immediately close  
217 the clamp attached to the rubber tube, then start the distillation with steam to obtain 80  
218 to 100 mL of the distillate. Remove the absorption flask from the lower end of the  
219 condenser tube, rinsing the end part with a small quantity of water, and titrate the  
220 distillate with 0.025 mol/L sulfuric acid VS until the color of the solution changes  
221 from green through pale grayish blue to pale grayish red-purple. Perform a blank  
222 determination in the same manner, and make any necessary correction.

223

224 Each mL of 0.025 mol/L sulfuric acid VS = 0.700 mg of  $N$

225

226

## 227 REAGENTS

228 **2-Pyrrolidone**  $C_4H_7NO$  Clear, colorless to pale yellow liquid or white to pale yellow  
229 crystalline masses. It is odorless.

230 *Refractive Index*  $n_D^{20}$ : 1.485~1.490

231 Congealing Point: 22~26°C

232 *Purity* – Weigh accurately about 1.0 g of 2-pyrrolidone, dissolve in methanol to make  
233 exactly 10 mL, and use this solution as the sample solution. Perform the test with  
234 1.0 µL of the sample solution as directed under the Gas Chromatography according  
235 to the following conditions. Determine each peak area of the solutions by the  
236 automatic integration method, and calculate the amount of 2-pyrrolidone by the area  
237 percentage method: it is not less than 98.0%.

238

239 *Operating conditions*

240 Detector: A hydrogen flame-ionization detector.

241 Column: A hollow, capillary glass column about 0.53 mm in inside diameter and  
242 about 30 m in length, having an about 1.0 µm layer of polyethylene glycol 20 M  
243 for gas chromatography on the inner side.

244 Column temperature: Maintain the temperature at 80°C for 1 minute, then raise at  
245 the rate of 10°C per minute to 190°C, and hold constant to the temperature for 20  
246 minutes.

247 Temperature of sample vaporization chamber: A constant temperature of about  
248 200°C. Carrier gas: Helium

249 Flow rate: Adjust the flow rate so that the retention time of 2-pyrrolidone is about 10  
250 minutes.

251 Time span of measurement: About twice as long as the retention time of  
252 2-pyrrolidone.

253 Split ratio: 1.20

254 Water: Not more than 0.2 %

255

256 **Acetaldehyde ammonia trimer trihydrate** (C<sub>2</sub>H<sub>5</sub>N)<sub>3</sub>·3H<sub>2</sub>O

257 *Purity* –Not less than 95.0%