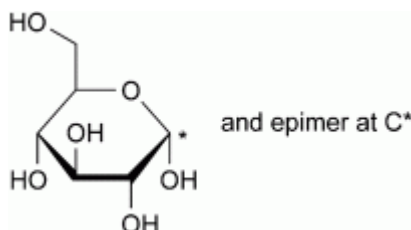
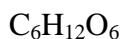


E-56 GLUCOSE, ANHYDROUS
Stage 4rev



M_r 180.2



DEFINITION

Glucose anhydrous is (+)-D-glucopyranose and is derived from starch.

Content: 97.5 per cent to 102.0 per cent (anhydrous substance), determined by the LC described in the assay.

IDENTIFICATION

A. Infrared absorption spectrophotometry.

Record the infrared absorption spectrum of anhydrous glucose and compare with the Reference Spectrum or the spectrum obtained with the Reference Standard: the transmission minima correspond in position and relative size.

B. Examine the chromatograms obtained in the assay.

The principal peak in the chromatogram obtained with the test solution is similar in retention time and size to the principal peak in the chromatogram obtained with reference solution (a).

C. Water (see tests).

TESTS

Appearance of solution. The test solution is clear (its clarity is the same as that of water or its opalescence is not more pronounced than that of reference suspension I) and not more intensely coloured than the reference solution.

Test solution: Dissolve 10.0 g in 15 mL of *water* using a bath of boiling water. Allow to cool.

Primary solutions:

- *Ferric chloride primary solution:* a 45.0 g/l solution of ferric chloride ($FeCl_3, 6H_2O$).

- *Cobalt chloride primary solution:* a 59.5 g/l solution of cobalt chloride ($CoCl_2, 6H_2O$).

- *Copper sulfate primary solution:* a 62.4 g/l solution of copper sulfate ($CuSO_4, 5H_2O$).

45

46 *Reference solution:* to 2.5 mL of cobalt chloride primary solution, 6.0 mL of ferric
47 chloride primary solution and 1.0 mL of copper sulfate primary solution, add
48 hydrochloric acid (10 g/l HCl) to make 1000.0 mL.

49

50 **Conductivity:** maximum $20 \mu\text{S}\cdot\text{cm}^{-1}$ at 25°C.

51 Dissolve 20.0 g in *carbon dioxide-free water* prepared from *distilled water* and dilute
52 to 100.0 mL with the same solvent. Measure the conductivity of the solution while
53 gently stirring with a magnetic stirrer.

54

55 **Related substances.** Liquid chromatography.

56 *Test solution.* Dissolve 0.300 g of the substance to be examined in *water*, and dilute to
57 10.0 mL with the same solvent.

58

59 *Reference solution (a).* Dissolve 0.330 g of *glucose monohydrate CRS* in *water* and
60 dilute to 10.0 mL with the same solvent.

61

62 *Reference solution (b).* Dilute 1.0 mL of the test solution to 250.0 mL with *water*.

63

64 *Reference solution (c).* Dilute 25.0 mL of reference solution (b) to 200.0 mL with
65 *water*.

66

67 *Reference solution (d).* Dissolve 5 mg of *maltose* (impurity A), 5 mg of *maltotriose*
68 (impurity C) and 5 mg of *fructose* (impurity D) in *water* and dilute to 50.0 mL with
69 *water*.

70

71 *Column:*

72 - size: $l = 0.3 \text{ m}$, $\varnothing = 7.8 \text{ mm}$;

73 - stationary phase: strong cation-exchange resin (calcium form) ($9\mu\text{m}$)¹;

74 - temperature: $85 \pm 1^\circ\text{C}$.

75

76 *Mobile phase:* degassed water.

77

78 *Flow rate:* 0.3 mL/min.

79

80 *Detection:* refractometer maintained at a constant temperature (40 °C for example).

81

82 *Injection:* 20 μl of the test solution and reference solutions (b), (c) and (d).

83

84 *Run time:* 1.5 times the retention time of glucose.

85

86 *Relative retention* with reference to glucose (retention time = about 21 min):
87 impurity C = about 0.7; impurities A and B = about 0.8; impurity D = about 1.3.

88

89 *System suitability* : reference solution (d) :

90 - *resolution* : minimum 1.3 between the peaks due to impurities C and A.

91

¹ Aminex HPX-87C from Biorad is suitable.

92 *Limits:*

93 - *sum of impurities A and B:* not more than the area of the principal peak in the
94 chromatogram obtained with reference solution (b) (0.4 per cent),

95 - *impurity C:* not more than 0.5 times the area of the principal peak in the
96 chromatogram obtained with reference solution (b) (0.2 per cent),

97 - *impurity D:* not more than 3 times the area of the principal peak in the
98 chromatogram obtained with reference solution (c) (0.15 per cent),

99 - *unspecified impurities:* for each impurity, not more than twice the area of the
100 principal peak in the chromatogram obtained with reference solution (c) (0.10 per
101 cent),

102 - *total:* not more than 1.25 times the area of the principal peak in the chromatogram
103 obtained with reference solution (b) (0.5 per cent),

104 - *disregard limit:* area of the principal peak in the chromatogram obtained with
105 reference solution (c) (0.05 per cent).

106

107 **Dextrin.** Reflux 1 g of the substance to be examined finely powdered with 20 mL of
108 ethanol (96 per cent): it dissolves completely.

109

110 **Soluble starch, sulphite:** maximum 15 ppm.

111 Dissolve 10.0 g in 15 mL of *water* using a bath of boiling water. Allow to cool and
112 add 50 μ l of 0.1 N iodine: the solution is yellow.

113

114 **Water:** maximum 1.0 per cent, determined on 0.50 g by the semi-micro determination
115 of water.

116

117 ASSAY

118

119 Liquid chromatography as described in the test for related substances with the
120 following modification.

121

122 *Injection:* test solution and reference solution (a).

123

124 Calculate the percentage content of $C_6H_{12}O_6$ from the areas of the peaks and the
125 assigned content of *glucose monohydrate CRS*.

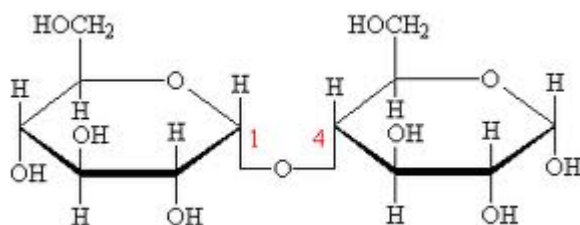
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127 IMPURITIES

128

129 A. Maltose

130



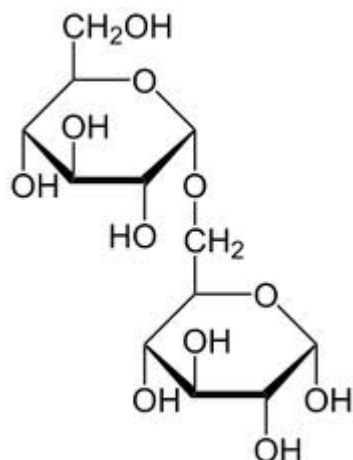
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134 B. Isomaltose

135



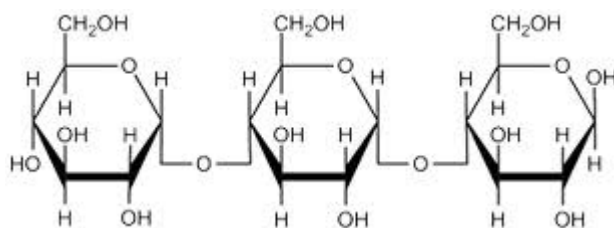
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139 C. Maltotriose

140

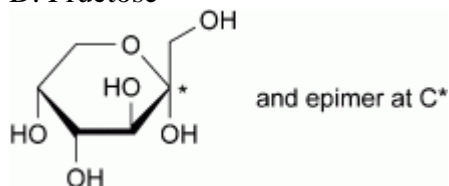


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142

143

144 D. Fructose



145

146

147

148

REAGENTS

149 **Hydrazine sulfate solution.** Dissolve 1.0 g of hydrazine sulfate in water and dilute to
150 100.0 mL with the same solvent. Allow to stand for 4-6 h.

151

152 **Hexamethylenetetramine solution.** In a 100 mL ground-glass-stoppered flask,
153 dissolve 2.5 g of hexamethylenetetramine in 25.0 mL of water.

154

155 **Primary opalescent suspension (formazin suspension).** To the
156 hexamethylenetetramine solution in the flask add 25.0 mL of the hydrazine sulfate
157 solution. Mix and allow to stand for 24 h. This suspension is stable for 2 months,
158 provided it is stored in a glass container free from surface defects. The suspension
159 must not adhere to the glass and must be well mixed before use.

160

161 **Standard of opalescence.** Dilute 15.0 mL of the primary opalescent suspension to
162 1000.0 mL with water. This suspension is freshly prepared and may be stored for up
163 to 24 h.

164

165 **Reference suspension I.** To 5.0 mL of standard of opalescence add 95.0 mL of water.
166 Mix and shake before use.

167

168 **Cation exchange resin (calcium form), strong.**

169 A resin in calcium form with sulfonic acid groups attached to a polymer lattice
170 consisting of polystyrene cross-linked with 8 per cent of divinylbenzene. The particle
171 size is specified after the name of the reagent in the tests where it is used.

172

173 **Fructose.** $C_6H_{12}O_6$. (M_r 180.2). [57-48-7].

174

175 **Maltose monohydrate.** $C_{12}H_{22}O_{11}$, H_2O . (M_r 360.3). [6363-53-7].

176

177 **Maltotriose.** $C_{18}H_{32}O_{16}$. (M_r 504.4). [1109-28-0].

178