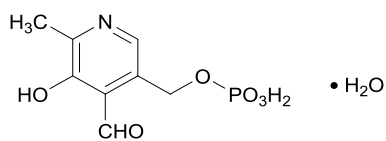


1 Pyridoxal Phosphate Hydrate

2 ピリドキサルリン酸エステル水和物



3
4 $C_8H_{10}NO_6P \cdot H_2O$: 265.16

5 (4-Formyl-5-hydroxy-6-methylpyridin-3-yl)methyl

6 dihydrogenphosphate monohydrate

7 [41468-25-1]

8

9 Pyridoxal Phosphate Hydrate contains not less than
10 98.0% and not more than 101.0% of pyridoxal phosphate
11 ($C_8H_{10}NO_6P$: 247.14), calculated on the anhydrous basis.

12 **Description** Pyridoxal Phosphate Hydrate occurs as a pale yel-
13 low-white to light yellow crystalline powder.

14 It is slightly soluble in water, and practically insoluble in etha-
15 nol (99.5).

16 It dissolves in dilute hydrochloric acid and in sodium hydroxide
17 TS.

18 The pH of a solution prepared by dissolving 0.1 g of Pyridoxal
19 Phosphate Hydrate in 200 mL of water is between 3.0 and 3.5.

20 Pyridoxal Phosphate Hydrate is colored to light red by light.

21 **Identification (1)** Determine the absorption spectrum of a so-
22 lution of Pyridoxal Phosphate Hydrate in phosphate buffer solu-
23 tion (pH 6.8) (1 in 50,000) as directed under Ultraviolet-visible
24 Spectrophotometry <2.24>, and compare the spectrum with the
25 Reference Spectrum or the spectrum of a solution of Pyridoxal
26 Phosphate RS prepared in the same manner as the sample solution:
27 both spectra exhibit similar intensities of absorption at the same
28 wavelengths.

29 **(2)** Determine the infrared absorption spectrum of Pyridoxal
30 Phosphate Hydrate as directed in the potassium bromide disk
31 method under Infrared Spectrophotometry <2.25>, and compare
32 the spectrum with the Reference Spectrum or the spectrum of Pyr-
33 idoxal Phosphate RS: both spectra exhibit similar intensities of ab-
34 sorption at the same wave numbers.

35 **Purity (1)** Heavy metals <1.07>—Proceed with 4.0 g of Pyri-
36 doxal Phosphate Hydrate according to Method 2, and perform the
37 test. Prepare the control solution with 2.0 mL of Standard Lead
38 Solution (not more than 5 ppm).

39 **(2)** Arsenic <1.11>—Dissolve 1.0 g of Pyridoxal Phosphate
40 Hydrate in 5 mL of dilute hydrochloric acid. Use this solution as
41 the test solution, and perform the test (not more than 2 ppm).

42 **(3)** Free phosphoric acid—Weigh accurately about 0.1 g of
43 Pyridoxal Phosphate Hydrate, dissolve in water to make exactly
44 100 mL, and use this solution as the sample solution. Pipet 5 mL
45 each of the sample solution and Standard Phosphoric Acid Solu-

46 tion, to each add 2.5 mL of hexaammonium heptamolybdate-sul-
47 furic acid TS and 1 mL of 1-amino-2-naphthol-4-sulfonic acid TS,
48 and shake. Add water to make exactly 25 mL, and allow to stand
49 at $20 \pm 1^\circ C$ for 30 minutes. Perform the test with these solutions
50 as directed under Ultraviolet-visible Spectrophotometry <2.24>,
51 using a solution prepared with 5 mL of water in the same manner
52 as the blank. Determine the absorbances, A_T and A_S , at 740 nm of
53 each solution from the sample solution and Standard Phosphoric
54 Acid Solution: the amount of free phosphoric acid is not more than
55 0.5%.

$$56 \quad \text{Content (\% of free phosphoric acid (H}_3\text{PO}_4\text{))}$$

$$57 \quad = 1/M \times A_T/A_S \times 258.0$$

58 M : Amount (mg) of Pyridoxal Phosphate Hydrate taken, calcu-
59 lated on the anhydrous basis

60 **(4)** Related substances—Dissolve 50 mg of Pyridoxal Phos-
61 phate Hydrate in 20 mL of the mobile phase, and use this solution
62 as the sample solution. Pipet 1 mL of the sample solution, add the
63 mobile phase to make exactly 100 mL, and use this solution as the
64 standard solution. Perform the test with exactly 5 μL each of the
65 sample solution and standard solution as directed under Liquid
66 Chromatography <2.01> according to the following conditions.
67 Determine each peak area by the automatic integration method:
68 the area of the peak other than pyridoxal phosphate from the sam-
69 ple solution is not larger than the peak area of pyridoxal phosphate
70 from the standard solution, and the total area of the peaks other
71 than pyridoxal phosphate from the sample solution is not larger
72 than 2 times the peak area of pyridoxal phosphate from the stand-
73 ard solution.

74 **Operating conditions**—

75 **Detector:** An ultraviolet absorption photometer (wavelength:
76 254 nm).

77 **Column:** A stainless steel column 4.6 mm in inside diameter and
78 25 cm in length, packed with octadecylsilanized silica gel for
79 liquid chromatography (5 μm in particle diameter).

80 **Column temperature:** A constant temperature of about $30^\circ C$.

81 **Mobile phase:** Dissolve 18.15 g of potassium dihydrogen
82 phosphate and 28.38 g of anhydrous disodium hydrogenphosphate
83 in water to make 5 L.

84 **Flow rate:** Adjust so that the retention time of pyridoxal
85 phosphate is about 6 minutes.

86 **Time span of measurement:** About 2.5 times as long as the
87 retention time of pyridoxal phosphate, beginning after the solvent
88 peak.

89 **System suitability**—

90 **Test for required detectability:** Pipet 2 mL of the standard
91 solution, and add the mobile phase to make exactly 20 mL.
92 Confirm that the peak area of pyridoxal phosphate obtained with
93 5 μL of this solution is equivalent to 7 to 13% of that of pyridoxal
94 phosphate obtained with 5 μL of the standard solution.

95 **System performance:** When the procedure is run with 5 μL of
96 the standard solution under the above operating conditions, the
97 number of theoretical plates and the symmetry factor of the peak

98 of pyridoxal phosphate are not less than 3000 and not more than
99 1.5, respectively.

100 System repeatability: When the test is repeated 6 times with 5
101 μL of the standard solution under the above operating conditions,
102 the relative standard deviation of the peak area of pyridoxal
103 phosphate is not more than 2.0%.

104 **Water** <2.48> 6.0 – 9.0% (0.1 g, volumetric titration, direct ti-
105 tration. Use a solution prepared by dissolving 50 g of imidazole
106 for water determination in 100 mL of the dissolving solution in-
107 stead of methanol for water determination).

108 Dissolving solution: A solution containing 80% of 1-methoxy-
109 2-propanol, 18% of ethanol (99.5), 1% of imidazole and 1% of
110 imidazole hydrobromide.

111 **Assay** Weigh accurately about 45 mg each of Pyridoxal Phos-
112 phate Hydrate and Pyridoxal Phosphate RS (separately determine
113 the water <2.48> in the same manner as Pyridoxal Phosphate Hy-
114 drate), and dissolve each in phosphate buffer solution (pH 6.8) to
115 make exactly 250 mL. Pipet 10 mL each of these solutions, add
116 phosphate buffer solution (pH 6.8) to make exactly 100 mL, and
117 use these solutions as the sample solution and the standard solu-
118 tion. Determine the absorbances, A_T and A_S , of the sample solution
119 and standard solution at 388 nm as directed under Ultraviolet-vis-
120 ible Spectrophotometry <2.24> using phosphate buffer solution
121 (pH 6.8) as the blank.

122 Amount (mg) of pyridoxal phosphate ($\text{C}_8\text{H}_{10}\text{NO}_6\text{P}$)
123
$$= M_S \times A_T / A_S$$

124 M_S : Amount (mg) of Pyridoxal Phosphate RS taken, calculated
125 on the anhydrous basis

126 **Containers and storage** Containers—Well-closed containers.
127 Storage—Light-resistant.

128 **Add the following to 9.01 Reference Standards**
129 **(1):**

130 **Pyridoxal Phosphate RS**

131 **Add the following to 9.41 Reagents, Test So-**
132 **lutions:**

133 **Imidazole hydrobromide** $\text{C}_3\text{H}_4\text{N}_2\cdot\text{HBr}$ White to pale yel-
134 low crystals. Melting point: about 221°C .

135

136