

1 Hydroxyethylcellulose

2 ヒドロキシエチルセルロース
3 [9004-62-0]

4
5 This monograph is harmonized with the European Phar-
6 macopoeia and the U. S. Pharmacopoeia.

7 The parts of the text that are not harmonized among the
8 targeted texts for the harmonization are marked with sym-
9 bols (◆ ◆), and the texts that are uniquely specified by the
10 JP other than the targeted texts for the harmonization are
11 marked with symbols (◇ ◇).

12 Hydroxyethylcellulose is partly *O*-(2-
13 hydroxyethylated) cellulose.

14 It contains not less than 30.0% and not more than
15 70.0% of hydroxyethoxy group (-OC₂H₄OH: 61.06),
16 calculated on the dried basis.

17 It may contain suitable pH-stabilizers such as phos-
18 phates.

19 ◆ The viscosity is shown in millipascal second
20 (mPa·s) on the label.◆

21 ◆**Description** Hydroxyethylcellulose occurs as a white to
22 yellowish white, powder or grains.

23 It is practically insoluble in ethanol (95).

24 It forms a viscous liquid upon addition of water.

25 It is hygroscopic.◆

26 **Identification (1)** Determine the infrared absorption
27 spectrum of Hydroxyethylcellulose as directed in the ATR
28 method under Infrared Spectrophotometry <2.25> and com-
29 pare the spectrum with the spectrum of Hydroxyethylcellu-
30 lose RS for Identification: both spectra exhibit similar in-
31 tensities of absorption at the same wave numbers.

32 **(2)** Disperse 1.0 g of Hydroxyethylcellulose in 50 mL
33 of freshly boiled and cooled water. After 10 minutes, add
34 freshly boiled and cooled water to make 100 mL, stir to
35 dissolve completely, and use this solution as the sample so-
36 lution. Boil 10 mL of the sample solution: the solution is
37 clear.

38 ◆**Viscosity <2.53>** Weigh exactly a quantity of Hydroxy-
39 ethylcellulose, equivalent to 10.00 g calculated on the dried
40 basis, add 400 mL of water, stir to dissolve, and add water
41 to make exactly 500.0 g. Remove air bubbles, and use this
42 solution as the sample solution. Perform the test with the
43 sample solution according to Method 2 at 20 ± 0.1°C, us-
44 ing a beaker with an inside diameter of not less than 70 mm
45 and a single cylinder-type rotational viscometer, according
46 to the following operating conditions: not less than 75%
47 and not more than 140% of the labeled viscosity.

48 *Operating conditions* —

49 Apparatus: Brookfield type viscometer LV and RV
50 model.

51 Rotor No., rotation frequency and calculation multiplier:
52 According to the following table, depending on the labeled
53 viscosity.

| Labeled viscosity (mPa·s) | Model | Rotor No. | Rotation frequency /min | Calculation multiplier |
|--------------------------------------|-------|--------------|-------------------------------|---------------------------|
| less than 200 | LV | 1 | 30 | 2 |
| Not less than 200 and less than 4000 | LV | 3 | 30 | 40 |
| " 4000 " 10,000 | LV | 4 | 30 | 200 |
| " 10,000 " 50,000 | RV | 6 | 20 | 500 |
| " 50,000 | RV | 7 | 20 | 2000 |

55
56 Procedure of apparatus: Read a value after 2 minutes of
57 rotation, and stop the rotation for at least 2 minutes. Repeat
58 this procedure more two times, and average the three
59 observed values.◆

60 **pH <2.54>** The pH of the sample solution obtained in the
61 Identification (2) is between 5.5 and 8.5.

62 **Purity (1) Chloride**—To 1 mL of the sample solution
63 obtained in the Identification (2) add water to make 30 mL,
64 and use this solution as the sample solution. Separately, to
65 10 mL of Standard Chloride Solution add 5 mL of water,
66 and use this solution as the control solution. To 15 mL each
67 of the sample solution and the control solution add 1 mL of
68 2 mol/L nitric acid TS, transfer to test tubes containing 1
69 mL of a solution of silver nitrate (17 in 1000), allow to
70 stand for 5 minutes protecting from light, and compare the
71 opalescence developed in the both solutions against a black
72 background by viewing transversely: the opalescence de-
73 veloped in the sample solution is not more intense than that
74 of the control solution (not more than 1.0%).

75 **(2) Nitrate**—Dissolve 0.50 g of Hydroxyethylcellulose
76 in the diluting solution to make exactly 100 mL, and use
77 this solution as the sample solution. Separately, dissolve
78 0.8154 g of potassium nitrate in the diluting solution to
79 make 1000 mL, and use this solution as the standard nitrate
80 stock solution. If the viscosity of Hydroxyethylcellulose is
81 not more than 1000 mPa·s, pipet 10 mL, 20 mL and 40 mL
82 of the standard nitrate stock solution, add the diluting solu-
83 tion to each to make exactly 100 mL, and use these solu-
84 tions as the standard solutions. If the viscosity of Hydrox-
85 yethylcellulose is more than 1000 mPa·s, pipet 1 mL, 2 mL
86 and 4 mL of the standard nitrate stock solution, add the di-
87 luting solution to each to make exactly 100 mL, and use
88 these solutions as the standard solutions. Perform the test
89 with the sample solution and standard solutions using a ni-
90 trate-selective electrode as an indicator electrode, a silver-
91 silver chloride electrode as a reference electrode and di-
92 luted ammonium sulfate TS (1 in 30) as reference electro-
93 lyte. Calculate the concentration of nitrates in the sample
94 solution using a calibration curve obtained from the poten-
95 tial differences of the standard solutions: not more than

96 3.0%, calculated on the dried basis, if Hydroxyethylcellu-
 97 lose has a viscosity of not more than 1000 mPa·s, and not
 98 more than 0.2%, calculated on the dried basis, if Hydroxy-
 99 ethylcellulose has a viscosity of more than 1000 mPa·s.
 100 Prepare the solutions before use.

101 Diluting solution: To a mixture of 50 mL of 1 mol/L sul-
 102 furic acid TS and 800 mL of water add 135 g of potassium
 103 dihydrogen phosphate, and add water to make 1000 mL. To
 104 this solution add water to make exactly 25 times the initial
 105 volume.

106 In order to determine the applicable limit, determine the
 107 viscosity using the following procedure.

108 Introduce a quantity of Hydroxyethylcellulose, equiva-
 109 lent to 2.00 g calculated on the dried basis, into 50 g of wa-
 110 ter, stir, add water to make 100 g, and stir to dissolve com-
 111 pletely. Determine the viscosity using a rotating viscometer
 112 at 25°C and at a shear rate of 100 s⁻¹ for substances with an
 113 expected viscosity less than 100 mPa·s, at a shear rate of
 114 10 s⁻¹ for substances with an expected viscosity not less
 115 than 100 mPa·s and not more than 20,000 mPa·s, and at a
 116 shear rate of 1 s⁻¹ for substances with an expected viscosity
 117 more than 20,000 mPa·s. If it is impossible to obtain a shear
 118 rate of exactly 10 s⁻¹ or 100 s⁻¹ respectively, use a rate
 119 slightly higher and a rate slightly lower and interpolate.

120 ◇(3) Heavy metal <1.07>—Proceed with 1.0 g of Hy-
 121 droxyethylcellulose according to Method 2, and perform
 122 the test. Prepare the control solution with 2.0 mL of Stand-
 123 ard Lead Solution (not more than 20 ppm). ◇

124 (4) Aldehydes—Introduce 1.0 g of Hydroxyethylcellu-
 125 lose into a glass-stoppered test tube, add 10 mL of ethanol
 126 (99.5), stopper the tube tightly, and stir for 30 minutes.
 127 Centrifuge, and use the supernatant liquid as the sample so-
 128 lution. Use Standard Glyoxal Solution as the control solu-
 129 tion. Pipet 2 mL each of the sample solution and the control
 130 solution, to each add 5 mL of a solution prepared by dis-
 131 solving 4 g of 3-methyl-2-benzothiazolonehydrazone hy-
 132 drochloride monohydrate in diluted acetic acid (100) (4 in
 133 5) to make 1000 mL, shake to become uniform, and allow
 134 to stand for 2 hours. Compare the color of these solutions:
 135 the sample solution is not more intensely colored than the
 136 control solution (not more than 20 ppm).

137 **Loss on drying** <2.41> Not more than 10.0% (1 g, 105°C,
 138 3 hours).

139 **Residue on Ignition** <2.44> Not more than 4.0% if the
 140 viscosity of Hydroxyethylcellulose is not more than 1000
 141 mPa·s, and not more than 1.0% if the viscosity of Hydrox-
 142 yethylcellulose is more than 1000 mPa·s (1 g). In order to
 143 determine the applicable limit, determine the viscosity ac-
 144 cording to the method in the Purity (2).

145 **Assay** Weigh accurately about 30 mg of Hydroxyethyl-
 146 cellulose, transfer to a 5-mL pressure-tight serum vial, add

147 exactly 60 mg of adipic acid, 2 mL of the internal standard
 148 solution and 1 mL of hydroiodic acid, seal the vial imme-
 149 diately with a septum coated with fluororesin and an alu-
 150 minium cap, and weigh accurately the vial. Place the vial in
 151 an oven or heat in a suitable heater with continuous stirring,
 152 maintaining the internal temperature of about 165 ± 2°C
 153 for 2.5 hours. Allow to cool and weigh accurately the vial.
 154 If the difference of the mass between before heating and
 155 after heating is more than 10 mg, prepare a new sample so-
 156 lution. If the difference of the mass between before heating
 157 and after heating is not more than 10 mg, after phase sepa-
 158 ration, pierce through the septum of the vial with a cooled
 159 syringe, and withdraw a sufficient volume of the upper
 160 phase as the sample solution. Separately, place exactly 60
 161 mg of adipic acid, 2 mL of the internal standard solution
 162 and 1 mL of hydroiodic acid in another serum vial, and seal
 163 immediately. Weigh accurately the vial, inject 55 µL of io-
 164 doethane for assay through the septum in the vial, and
 165 weigh again accurately. Shake thoroughly, after phase sep-
 166 aration, pierce through the septum of the vial with a cooled
 167 syringe, and withdraw a sufficient volume of the upper
 168 phase as the standard solution. Perform the test as directed
 169 under Gas Chromatography <2.02> with 1 µL each of the
 170 sample solution and standard solution according to the fol-
 171 lowing conditions, calculate the ratios, Q_T and Q_S , of the
 172 peak area of iodoethane to that of the internal standard.

$$173 \quad \text{Amount (\%)} \text{ of hydroethoxy group } (-OC_2H_4OH) \\ 174 \quad = M_S / M_T \times Q_T / Q_S \times 39.15$$

175 M_S : Amount (mg) of iodoethane for assay taken

176 M_T : Amount (mg) of Hydroxyethylcellulose taken, cal-
 177 culated on the dried basis

178 *Internal standard solution*—A solution of *n*-octane in *o*-
 179 xylene (1 in 200).

180 *Operating conditions*—

181 Detector: A hydrogen flame-ionization detector.

182 Column: A fused silica column 0.53 mm in inside
 183 diameter and 30 m in length, coated with
 184 dimethylpolysiloxane for gas chromatography in 3 µm
 185 thickness.

186 Column temperature: Maintain the temperature at 50°C
 187 for 3 minutes, raise the temperature to 100°C at a rate of
 188 10°C per minute, then to 250°C at a rate of 35°C per minute,
 189 and maintain at 250°C for 8 minutes.

190 Injection port temperature: A constant temperature of
 191 about 250°C.

192 Detector temperature: A constant temperature of about
 193 280°C.

194 Carrier gas: Helium.

195 Flow rate: 4.2 mL per minutes (the retention time of the
 196 internal standard is about 10 minutes).

197 Split ratio: 1:40.

198 *System suitability*—
 199 System performance: When the procedure is run with 1
 200 μL of the standard solution under the above operating
 201 conditions, iodoethane and the internal standard are eluted
 202 in this order with the relative retention time of iodoethane
 203 to that of the internal standard being about 0.6 and the
 204 resolution between these peaks being not less than 5.0.

205 System repeatability: When the test is repeated 6 times
 206 with 1 μL of the standard solution under the above
 207 operating conditions, the relative standard deviation of the
 208 ratio of the peak area of iodoethane to that of the internal
 209 standard is not more than 2.0%.

210 **♦Containers and storage** Containers—Tight containers.
 211 ♦

212 **Add the following to 9.01 Reference**
 213 **Standards (1):**

214 **Hydroxyethylcellulose RS for Identification**

215 **Add the following to 9.22 Standard So-**
 216 **lutions:**

217 **Standard Chloride Solution** Pipet 10 mL of Standard
 218 Chloride Stock Solution, add water to make exactly 1000
 219 mL. Prepare before use. Each mL of this solution contains
 220 5 μg of chloride (Cl).

221 **Standard Chloride Stock Solution** Weigh accurately
 222 0.842 g of sodium chloride, previously dried at 130°C for
 223 2 hours, and dissolve in water to make exactly 1000 mL.

224 **Standard Glyoxal Solution** Dilute Standard Glyoxal
 225 Stock Solution to 10 times with ethanol (99.5). Prepare be-
 226 fore use. Each mL of this solution contains 0.002 mg of
 227 glyoxal ($\text{C}_2\text{H}_2\text{O}_2$).

228 **Standard Glyoxal Stock Solution** Transfer a quantity
 229 of 40% glyoxal TS, equivalent to 0.200 g of glyoxal, in a
 230 100-mL volumetric flask, and dilute to 100 mL with etha-
 231 nol (99.5). Dilute to 100-fold with ethanol (99.5) before use.
 232 Each mL of this solution contains 0.02 mg of glyoxal
 233 ($\text{C}_2\text{H}_2\text{O}_2$).

234 **Add the following to 9.41 Reagents,**
 235 **Test Solutions:**

236 **40% glyoxal TS** *Content:* 38 – 42%. Assay—Put
 237 1.000 g of 40% glyoxal TS in a glass-stoppered flask, add
 238 20 mL of a solution of hydroxylammonium chloride (7 in
 239 100) and 50 mL of water. Stopper tightly, allow to stand for
 240 30 minutes, titrate <2.50> with 1 mol/L sodium hydroxide
 241 VS (indicator: 1.0 mL of methyl red-methylene blue TS).
 242 Perform a blank determination in the same manner, and
 243 make any necessary correction.

244 Each mL of 1 mol/L sodium hydroxide VS
 245 = 29.02 mg of $\text{C}_2\text{H}_2\text{O}_2$

246 **Iodoethane for assay** $\text{C}_2\text{H}_5\text{I}$ Colorless to slightly
 247 yellow liquid, turning brown on exposure to air and light.
 248 Miscible with ethanol (95). Boiling point: about 72°C; Spe-
 249 cific gravity d_{20}^{20} : about 1.95.

250 *Refractive index* <2.45> n_D^{20} : about 1.513.

251 *Content:* not less than 99.0%. Assay—Proceed as di-
 252 rected in the Assay under isopropyl iodide for assay.

253 Each mL of 0.1 mol/L silver nitrate VS
 254 = 15.60 mg of $\text{C}_2\text{H}_5\text{I}$

255 *Storage*—Preserve in tight, light-resistant containers.

256 **3-Methyl-2-benzothiazolonehydrazone hydrochloride monohydrate** $\text{C}_8\text{H}_{10}\text{ClN}_3\text{S}\cdot\text{H}_2\text{O}$ A white to light
 257 yellow-white crystalline powder.

258 *Melting point* <2.60>: about 270°C (with decomposi-
 260 tion).

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