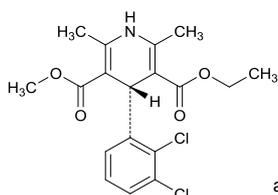


# 1 Felodipine

2 フェロジピン



3 and enantiomer

4  $C_{18}H_{19}Cl_2NO_4$ : 384.25

5 Ethyl methyl (4*RS*)-4-(2,3-dichlorophenyl)-2,6-dimethyl-1,4-

6 dihydropyridine-3,5-dicarboxylate

7 [72509-76-3]

8

9 Felodipine contains not less than 99.0% and not  
10 more than 101.0% of felodipine ( $C_{18}H_{19}Cl_2NO_4$ ),  
11 calculated on the dried basis.

12 **Description** Felodipine occurs as pale yellow-white to  
13 light yellow-white, crystals or crystalline powder.

14 It is freely soluble in methanol and in ethanol (99.5), and  
15 practically insoluble in water.

16 A solution of Felodipine in methanol (1 in 20) shows no  
17 optical rotation.

18 **Identification (1)** Determine the absorption spectrum  
19 of a solution of Felodipine in methanol (1 in 62,500) as di-  
20 rected under Ultraviolet-visible Spectrophotometry <2.24>,  
21 and compare the spectrum with the Reference Spectrum:  
22 both spectra exhibit similar intensities of absorption at the  
23 same wavelengths.

24 **(2)** Determine the infrared absorption spectrum of Fe-  
25 lodipine as directed in the potassium bromide disk method  
26 under Infrared Spectrophotometry <2.25>, and compare the  
27 spectrum with the Reference Spectrum: both spectra ex-  
28 hibit similar intensities of absorption at the same wave  
29 numbers.

30 **Purity (1)** Heavy metals—Being specified separately  
31 when the drug is granted approval based on the Law.

32 **(2)** Related substances—Dissolve 25 mg of Felodipine  
33 in 50 mL of the mobile phase, and use this solution as the  
34 sample solution. Pipet 1 mL of the sample solution, and add  
35 the mobile phase to make exactly 100 mL. Pipet 1 mL of  
36 this solution, add the mobile phase to make exactly 10 mL,  
37 and use this solution as the standard solution. Perform the  
38 test with 20  $\mu$ L each of the sample solution and standard  
39 solution as directed under Liquid Chromatography <2.01>  
40 according to the following conditions. Determine each  
41 peak area by the automatic integration method: the area of  
42 the peak other than felodipine, the related substance B, hav-  
43 ing the relative retention time of about 0.7 to felodipine,

44 and the related substance C, having the relative retention  
45 time of about 1.4, obtained from the sample solution is not  
46 larger than the peak area of felodipine from the standard  
47 solution. Furthermore, the total area of the peaks of related  
48 substances B and C is not larger than 10 times the peak area  
49 of felodipine from the standard solution, and the total area  
50 of the peaks other than felodipine and related substances  
51 mentioned above is not larger than 3 times the peak area of  
52 felodipine from the standard solution. For this calculation  
53 the peak area less than 1/5 times the peak area of felodipine  
54 from the standard solution is excluded.

55 **Operating conditions—**

56 **Detector:** An ultraviolet absorption photometer  
57 (wavelength: 254 nm).

58 **Column:** A stainless steel column 4.6 mm in inside  
59 diameter and 15 cm in length, packed with  
60 octadecylsilanized silica gel for liquid chromatography (5  
61  $\mu$ m in particle diameter).

62 **Column temperature:** A constant temperature of about  
63 25°C.

64 **Mobile phase:** Dissolve 3.2 g of sodium dihydrogen  
65 phosphate dihydrate in 400 mL of water, adjust to pH 3.0  
66 with phosphoric acid, and add 200 mL of methanol and 400  
67 mL of acetonitrile.

68 **Flow rate:** Adjust so that the retention time of felodipine  
69 is about 12 minutes.

70 **Time span of measurement:** About 2 times as long as the  
71 retention time of felodipine, beginning after the solvent peak.

72 **System suitability—**

73 **Test for required detectability:** When the procedure is  
74 run with 20  $\mu$ L of the standard solution under the above  
75 operating conditions, the SN ratio of the peak of felodipine  
76 is not less than 30.

77 **System performance:** Dissolve 25 mg of Felodipine in 50  
78 mL of the mobile phase. To 1 mL of this solution add the  
79 mobile phase to make 100 mL. To 1 mL of this solution add  
80 the mobile phase to make 10 mL. When the procedure is  
81 run with 20  $\mu$ L of this solution under the above operating  
82 conditions, the number of theoretical plates and the  
83 symmetry factor of the peak of felodipine are not less than  
84 5000 and not more than 1.5, respectively.

85 **System repeatability:** When the test is repeated 6 times  
86 with 20  $\mu$ L of the standard solution under the above  
87 operating conditions, the relative standard deviation of the  
88 peak area of felodipine is not more than 2.0%.

89 **Loss on drying <2.41>** Not more than 0.2% (1 g, 105°C,  
90 3 hours).

91 **Residue on ignition <2.44>** Not more than 0.1% (1 g).

92 **Assay** Weigh accurately about 0.16 g of Felodipine, dis-  
93 solve in 25 mL of *t*-butyl alcohol and 25 mL of diluted per-  
94 chloric acid (17 in 200), and titrate with 0.1 mol/L cerium

95 (IV) sulfate VS <2.50> (indicator: 50  $\mu$ L of 1,10-phenan-  
 96 throline TS) until the color of the solution changes from  
 97 orange to colorless. Perform a blank determination in the  
 98 same manner, and make any necessary correction.

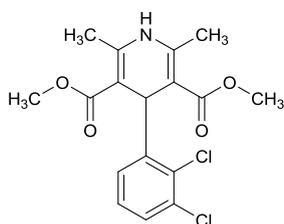
99 Each mL of 0.1 mol/L cerium sulfate (IV) VS  
 100 = 19.21 mg of  $C_{18}H_{19}Cl_2NO_4$

101 **Containers and storage** Containers—Well-closed con-  
 102 tainers.

### 103 Others

104 Related substance B:

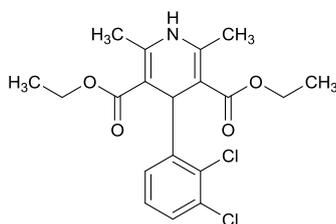
105 Dimethyl 4-(2,3-dichlorophenyl)-2,6-dimethyl-1,4-dihy-  
 106 dropyridine-3,5-dicarboxylate



107

108 Related substance C:

109 Diethyl 4-(2,3-dichlorophenyl)-2,6-dimethyl-1,4-dihydro-  
 110 pyridine-3,5-dicarboxylate



111

112

113 **Add the following to 9.21 Standard So-**  
 114 **lutions for Volumetric Analysis:**

#### 115 **Cerium (IV) Sulfate, 0.1 mol / L**

116 1000 mL of this solution contains 40.43 g of cerium (IV)  
 117 sulfate tetrahydrate ( $Ce(SO_4)_2 \cdot 4H_2O$ : 404.30).

118 *Preparation*—Dissolve 40.43 g of cerium sulfate (IV)  
 119 tetrahydrate in water to make 1000 mL, and standardize the  
 120 solution as follows:

121 *Standardization*—Weigh accurately about 0.2 g of so-  
 122 dium oxalate (standard reagent), previously dried at 150 to  
 123 200°C for 1 to 1.5 hours, and allowed to cool in a desicca-  
 124 tor (silica gel), and dissolve in 75 mL of water. Add a mix-  
 125 ture of 5 mL of water and 2 mL of sulfuric acid with stirring,  
 126 add 10 mL of hydrochloric acid. Warm to 70 – 75°C, and

127 titrate <2.50> the solution with 0.1 mol/L cerium (IV) sul-  
 128 fate VS until the solution shows a persistent slightly yellow  
 129 color, and calculate the molarity factor.

130 Each mL of 0.1 mol/L cerium sulfate (IV) VS  
 131 = 6.700 mg of  $Na_2C_2O_4$

132

133