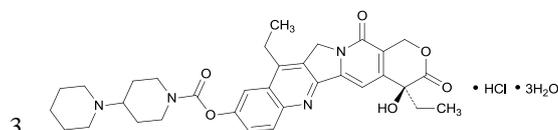


1 Irinotecan Hydrochloride Hydrate

2 イリノテカン塩酸塩水和物



4 $C_{33}H_{38}N_4O_6 \cdot HCl \cdot 3H_2O$: 677.18

5 (4S)-4,11-Diethyl-4-hydroxy-3,14-dioxo-3,4,12,14-tetrahydro-1H-
6 pyrano[3',4':6,7]indolizino[1,2-b]quinolin-9-yl [1,4'-bipiperidine]-1'-
7 carboxylate monohydrochloride trihydrate

8 [136572-09-3]

9

10 Irinotecan Hydrochloride Hydrate contains not
11 less than 99.0% and not more than 102.0% of iri-
12 notecan hydrochloride ($C_{33}H_{38}N_4O_6 \cdot HCl$: 623.14).

13 **Description** Irinotecan Hydrochloride Hydrate occurs
14 as pale yellow to light yellow, crystals or crystalline
15 powder.

16 It is sparingly soluble in methanol, and slightly soluble
17 in water and in ethanol (99.5).

18 It is gradually colored to yellow-brown and decom-
19 posed by light.

20 Melting point: about 255°C (with decomposition).

21 Irinotecan Hydrochloride Hydrate shows crystal poly-
22 morphism.

23 **Identification** (1) Determine the absorption spectrum
24 of a solution of Irinotecan Hydrochloride Hydrate in
25 methanol (1 in 100,000) as directed under Ultravio-
26 let-visible Spectrophotometry <2.24>, and compare the
27 spectrum with the Reference Spectrum: both spectra ex-
28 hibit similar intensities of absorption at the same wave-
29 lengths.

30 (2) Determine the infrared absorption spectrum of
31 Irinotecan Hydrochloride Hydrate as directed in the po-
32 tassium bromide disk method under Infrared Spectropho-
33 tometry <2.25>, and compare the spectrum with the Ref-
34 erence Spectrum: both spectra exhibit similar intensities
35 of absorption at the same wave numbers.

36 (3) To 1 g of Irinotecan Hydrochloride Hydrate add
37 50 mL of water, dissolve by warming, and cool: the solu-
38 tion responds to Qualitative Tests <1.09> (2) for chloride.

39 **Optical rotation** <2.49> $[\alpha]_D^{20}$: + 64 – + 69° (0.5 g cal-
40 culated on the anhydrous basis, water, heat, after cooling,
41 50 mL, 100 mm).

42 **pH** <2.54> Dissolve 1 g of Irinotecan Hydrochloride
43 Hydrate in 50 mL of water by warming, and cool: the pH
44 of this solution is between 3.5 and 4.5.

45 **Purity** (1) Heavy metals <1.07>—Proceed with 2.0 g
46 of Irinotecan Hydrochloride Hydrate according to Method
47 2, and perform the test. Prepare the control solution with
48 2.0 mL of Standard Lead Solution (not more than 10
49 ppm).

50 (2) Related substances—Dissolve 50 mg of Irinotecan
51 Hydrochloride Hydrate in a mixture of diluted 0.1 mol/L
52 potassium dihydrogen phosphate TS (1 in 10), methanol
53 and acetonitrile (6:4:3) and 1 mL of 1 mol/L hydrochloric
54 acid TS to make 20 mL, and use this solution as the sam-
55 ple solution. Pipet 1 mL of the sample solution, add a
56 mixture of diluted 0.1 mol/L potassium dihydrogen phos-
57 phate TS (1 in 10), methanol and acetonitrile (6:4:3) to
58 make exactly 100 mL, and use this solution as the stand-
59 ard solution. Perform the test with exactly 20 μL each of
60 the sample solution and standard solution as directed un-
61 der Liquid Chromatography <2.01> according to the fol-
62 lowing conditions. Determine each peak area by the au-
63 tomatic integration method: the peak areas of related sub-
64 stances A and B, having the relative retention time of
65 about 0.8 to irinotecan, and related substances C and D,
66 having the relative retention time of about 1.6, obtained
67 from the sample solution are not larger than 1/5 times the
68 peak area of irinotecan from the standard solution, and the
69 area of the peak other than irinotecan and the peaks men-
70 tioned above from the sample solution is not larger than
71 1/10 times the peak area of irinotecan from the standard
72 solution. Furthermore, the total area of the peaks other
73 than irinotecan from the sample solution is not larger than
74 4/5 times the peak area of irinotecan from the standard
75 solution.

76 **Operating conditions**—

77 Detector: An ultraviolet absorption photometer (wave-
78 length: 254 nm).

79 Column: A stainless steel column 4.6 mm in inside di-
80 ameter and 25 cm in length, packed with octadecylsi-
81 lanized silica gel for liquid chromatography (5 μm in par-
82 ticle diameter).

83 Column temperature: A constant temperature of about
84 40°C.

85 Mobile phase: Dissolve 1.22 g of sodium
86 1-decanesulfonate in a mixture of diluted 0.1 mol/L po-
87 tassium dihydrogen phosphate TS (1 in 10), methanol and
88 acetonitrile (6:4:3) to make 1000 mL.

89 Flow rate: Adjust so that the retention time of irinotec-
90 an is about 12 minutes.

91 Time span of measurement: About 3 times as long as
92 the retention time of irinotecan.

93 **System suitability**—

94 Test for required detectability: Pipet 1 mL of the stand-
95 ard solution, and add a mixture of diluted 0.1 mol/L po-
96 tassium dihydrogen phosphate TS (1 in 10), methanol and

97 acetonitrile (6:4:3) to make exactly 20 mL. Confirm that
 98 the peak area of irinotecan obtained with 20 μ L of this
 99 solution is equivalent to 3.5 to 6.5% of that with 20 μ L of
 100 the standard solution.

101 System performance: When the procedure is run with
 102 20 μ L of the standard solution under the above operating
 103 conditions, the number of theoretical plates and the sym-
 104 metry factor of the peak of irinotecan are not less than
 105 6000 and not more than 2.0, respectively.

106 System repeatability: When the test is repeated 6 times
 107 with 20 μ L of the standard solution under the above oper-
 108 ating conditions, the relative standard deviation of the
 109 peak area of irinotecan is not more than 2.0%.

110 (3) Optical isomer—Being specified separately when
 111 the drug is granted approval based on the Law.

112 **Water** <2.48> 7.5 – 9.5% (0.1 g, volumetric titration,
 113 direct titration).

114 **Residue on ignition** <2.44> Not more than 0.1% (1 g).

115 **Assay** Weigh accurately about 0.44 g of Irinotecan Hy-
 116 drochloride Hydrate, dissolve in 120 mL of a mixture of
 117 acetic anhydride and acetic acid (100) (7:3), and titrate
 118 <2.50> with 0.1 mol/L perchloric acid VS (potentiometric
 119 titration). Perform a blank determination in the same
 120 manner, and make any necessary correction.

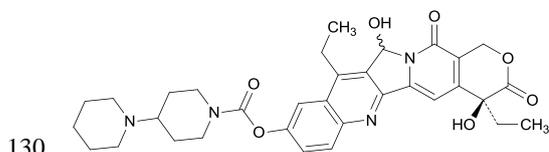
121 Each mL of 0.1 mol/L perchloric acid VS
 122 = 31.16 mg of $C_{33}H_{38}N_4O_6 \cdot HCl$

123 **Containers and storage** Containers—Tight containers.
 124 Storage—Light-resistant.

125 Others

126 Related substance A:

127 (4S)-4,11-Diethyl-4,12-dihydroxy-3,14-dioxo-3,4,12,14-
 128 -tetrahydro-1H-pyrano[3',4':6,7]indolizino[1,2-b]quinoli-
 129 ne-9-yl [1,4'-bipiperidine]-1'-carboxylate

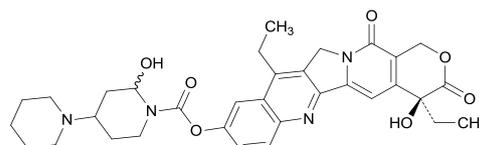


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132 Related substance B:

133 (4S)-4,11-Diethyl-4-hydroxy-3,14-dioxo-3,4,12,14-tetra-
 134 hy-
 135 dro-1H-pyrano[3',4':6,7]indolizino[1,2-b]quinoline-9-yl
 136 2'-hydroxy-[1,4'-bipiperidine]-1'-carboxylate

137

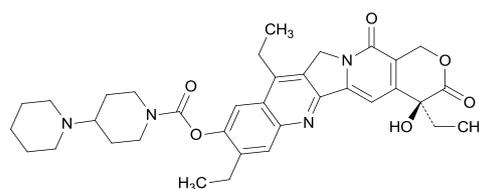


138

139 Related substance C:

140 (4S)-4,8,11-Triethyl-4-hydroxy-3,14-dioxo-3,4,12,14-tetra-
 141 hydro-
 142 dro-1H-pyrano[3',4':6,7]indolizino[1,2-b]quinoline-9-yl
 143 [1,4'-bipiperidine]-1'-carboxylate

144

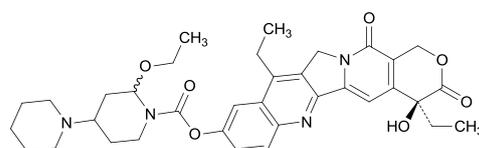


145

146 Related substance D:

147 (4S)-4,11-Diethyl-4-hydroxy-3,14-dioxo-3,4,12,14-tetra-
 148 hydro-
 149 dro-1H-pyrano[3',4':6,7]indolizino[1,2-b]quinoline-9-yl
 150 2'-ethoxy-[1,4'-bipiperidine]-1'-carboxylate

151



152