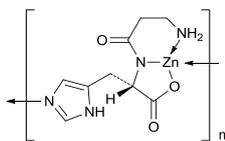


1 Polaprezinc

2 ポラプレジンク

3



4

5

6 $(C_9H_{12}N_4O_3Zn)_n$

7 *catena*-Poly[zinc- μ -[β -alanyl-L-histidinato(2-)-*N,N^V,O:N^T*]]

8 [107667-60-7]

9

10 Polaprezinc contains not less than 98.0% and not
11 more than 102.0% of polaprezinc ($C_9H_{12}N_4O_3Zn$:
12 289.60), and contains not less than 21.5% and not
13 more than 23.0% of zinc (Zn: 65.38), calculated on
14 the anhydrous basis.

15 **Description** Polaprezinc occurs as a white to pale yellow-
16 white crystalline powder.

17 It is practically insoluble in water, in methanol, and in
18 ethanol (99.5).

19 It dissolves in dilute hydrochloric acid, and in sodium
20 hydroxide TS.

21 **Identification** (1) To 2 mL of a solution of Polaprezinc
22 in 0.2 mol/L hydrochloric acid TS (1 in 1000) add 0.5
23 mL of a solution of sulfanilic acid in 1 mol/L hydrochloric
24 acid TS (1 in 200), 0.5 mL of a solution of sodium nitrite
25 (1 in 20) and 3 mL of sodium carbonate TS: a red color is
26 produced.

27 (2) A solution of Polaprezinc in 0.2 mol/L hydrochloric
28 acid TS (1 in 1000) responds to the Qualitative Tests
29 <1.09> for zinc salt.

30 (3) Determine the infrared absorption spectrum of
31 Polaprezinc as directed in the potassium bromide disk
32 method under Infrared Spectrophotometry <2.25>, and
33 compare the spectrum with the Reference Spectrum: both
34 spectra exhibit similar intensities of absorption at the
35 same wave numbers.

36 **Optical rotation** <2.49> $[\alpha]_D^{20}$: +8 ~ +9° (1 g calculated on the
37 anhydrous basis, 3 mol/L hydrochloric acid
38 TS, 50 mL, 100 mm).

39 **Purity** (1) Lead—Weigh accurately about 0.5 g of
40 Polaprezinc, dissolve in 3 mL of dilute nitric acid, add
41 water to make exactly 10 mL, and use this solution as the
42 sample solution. Separately, pipet 0.5 mL, 1.0 mL, 1.5 mL
43 and 2.0 mL of Standard Lead Solution, to each solution
44 add 3 mL of dilute nitric acid and water to make exactly
45 10 mL, and use these solutions as the standard solutions.

46 Perform the test with the sample solution and standard
47 solutions as directed under Atomic Absorption Spectro-
48 photometry <2.23> according to the following conditions,
49 and calculate the amount of lead in the sample solution
50 using a calibration curve obtained from the absorbances of
51 the standard solutions: not more than 10 ppm.

52 Gas:

53 Combustible gas—Acetylene.

54 Supporting gas—Air.

55 Lamp: Lead hollow-cathode lamp.

56 Wavelength: 283.3 nm.

57 (2) Related substances—Dissolve 50 mg of Polaprezinc
58 in 10 mL of 0.1 mol/L hydrochloric acid TS, add the
59 mobile phase to make 100 mL, and use this solution as the
60 sample solution. Pipet 1 mL of the sample solution, add
61 the mobile phase to make exactly 100 mL, and use this
62 solution as the standard solution. Perform the test with 10
63 μ L each of the sample solution and standard solution as
64 directed under Liquid Chromatography <2.01> according
65 to the following conditions. Determine each peak area by
66 the automatic integration method: the peak area of
67 L-histidine, having the relative retention time of about
68 0.38 to L-carnosine, obtained from the sample solution is
69 not larger than 1/5 times of the peak area of L-carnosine
70 from the standard solution, the area of the peak other than
71 L-carnosine and the peak mentioned above, from the sample
72 solution is not larger than 1/10 times of the peak area
73 of L-carnosine from the standard solution. Furthermore,
74 the total area of the peaks other than L-carnosine from the
75 sample solution is not larger than the peak area of
76 L-carnosine from the standard solution.

77 *Operating conditions*—

78 Detector, column, column temperature, mobile phase
79 and flow rate: Proceed as directed in the operating conditions
80 in the Assay.

81 Time span of measurement: About 4 times as long as
82 the retention time of L-carnosine, beginning after the solvent
83 peak.

84 *System suitability*—

85 Test for required detectability: Pipet 2 mL of the standard
86 solution, add the mobile phase to make exactly 20 mL.
87 Confirm that the peak area of L-carnosine obtained with
88 10 μ L of this solution is equivalent to 7 to 13% of that
89 with 10 μ L of the standard solution.

90 System performance: Dissolve 50 mg each of Polaprezinc
91 and L-histidine in 10 mL of 0.1 mol/L hydrochloric
92 acid TS, and add the mobile phase to make 100 mL. When
93 the procedure is run with 10 μ L of this solution under the
94 above operating conditions, L-histidine and L-carnosine
95 are eluted in this order with the resolution between these
96 peaks being not less than 12.

97 System repeatability: When the test is repeated 6 times
 98 with 10 μL of the standard solution under the above oper-
 99 ating conditions, the relative standard deviation of the
 100 peak area of L-carnosine is not more than 2.0%.

101 **Water** <2.48> Not more than 5.0% (0.2 g, volumetric
 102 titration, direct titration, stir for 30 minutes).

103 **Assay** (1) Weigh accurately about 25 mg of Polapre-
 104 zinc, dissolve in 5 mL of 0.1 mol/L hydrochloric acid TS,
 105 add the mobile phase to make exactly 100 mL, and use
 106 this solution as the sample solution. Separately, weigh
 107 accurately about 20 mg of L-Carnosine RS, previously
 108 dried at 105°C for 3 hours, dissolve in 5 mL of 0.1 mol/L
 109 hydrochloric acid TS, add the mobile phase to make ex-
 110 actly 100 mL, and use this solution as the standard solu-
 111 tion. Perform the test with 10 μL each of the sample solu-
 112 tion and standard solution as directed under Liquid Chro-
 113 matography <2.01> according to the following conditions.
 114 Determine the peak areas, A_T and A_S , of L-carnosine in
 115 each solution.

$$\begin{aligned} 116 & \text{Amount (mg) of polaprezinc (C}_9\text{H}_{12}\text{N}_4\text{O}_3\text{Zn)} \\ 117 & = M_S \times A_T / A_S \times 1.292 \end{aligned}$$

118 M_S : Amount (mg) of L-Carnosine RS

119 *Operating conditions*—

120 Detector: An ultraviolet absorption photometer (wave-
 121 length: 210 nm).

122 Column: A stainless steel column 4.6 mm in inside di-
 123 ameter and 15 cm in length, packed with octadecylsi-
 124 lanized silica gel for liquid chromatography (5 μm in par-
 125 ticle diameter).

126 Column temperature: A constant temperature of about
 127 45°C.

128 Mobile phase: Dissolve 1.4 g of potassium dihydrogen
 129 phosphate in 1000 mL of water, adjust to pH 3.5 with
 130 diluted phosphoric acid (1 in 100). Dissolve 2 g of sodium
 131 1-octane sulfonate in 900 mL of this solution, and add 100
 132 mL of acetonitrile for liquid chromatography.

133 Flow rate: Adjust so that the retention time of
 134 L-carnosine is about 15 minutes.

135 *System suitability*—

136 System performance: Dissolve 5 mg of L-histidine in 20
 137 mL of the standard solution. When the procedure is run
 138 with 10 μL of this solution under the above operating
 139 conditions, L-histidine and L-carnosine are eluted in this
 140 order with the resolution between these peaks being not
 141 less than 12.

142 System repeatability: When the test is repeated 6 times
 143 with 10 μL of the standard solution under the above oper-
 144 ating conditions, the relative standard deviation of the
 145 peak area of L-carnosine is not more than 1.0%.

146 (2) Zinc—Weigh accurately about 0.2 g of Polapre-
 147 zinc, dissolve in 3 mL of dilute hydrochloric acid TS, and
 148 add water to make exactly 100 mL. Pipet 25 mL of this
 149 solution, add 10 mL of ammonia-ammonium chloride
 150 buffer solution (pH 10.7), and titrate <2.50> with 0.01
 151 mol/L disodium dihydrogen ethylenediamine tetraacetate
 152 VS (indicator: 0.04 g of eriochrome black T-sodium chlo-
 153 ride indicator).

154 Each mL of 0.01 mol/L disodium dihydrogen ethylenedia-
 155 mine tetraacetate VS

156 = 0.6538 mg Zn

157 **Containers and storage** Containers—Tight containers.

158 **Change the following 9.01 Reference**
 159 **Standards (1) as follows:**

160 **L-Carnosine RS**

161