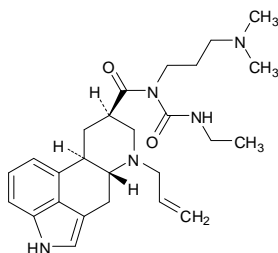


# 1 Cabergoline

2 カベルゴリン



3

4 C<sub>26</sub>H<sub>37</sub>N<sub>5</sub>O<sub>2</sub>: 451.60

5 (8*R*)-6-Allyl-*N*-[3-(dimethylamino)propyl]-*N*-

6 (ethylcarbamoyl)ergoline-8-carboxamide

7 [81409-90-7]

8

9 Cabergoline contains not less than 98.0% and not  
10 more than 102.0% of cabergoline (C<sub>26</sub>H<sub>37</sub>N<sub>5</sub>O<sub>2</sub>),  
11 calculated on the anhydrous basis.

12 **Description** Cabergoline occurs as a white crystalline  
13 powder.

14 It is very soluble in methanol, freely soluble in ethanol  
15 (95), and very slightly soluble in water.

16 It is gradually colored to yellow by light.

17 It shows crystal polymorphism.

18 **Identification** (1) Determine the absorption spectrum  
19 of a solution of Cabergoline in ethanol (95) (1 in 30,000) as  
20 directed under Ultraviolet-visible Spectrophotometry  
21 <2.24>, and compare the spectrum with the Reference Spec-  
22 trum or the spectrum of a solution of Cabergoline RS pre-  
23 pared in the same manner as the sample solution: both spec-  
24 tra exhibit similar intensities of absorption at the same  
25 wavelengths.

26 (2) Determine the infrared absorption spectrum of  
27 Cabergoline as directed in the potassium bromide disk  
28 method under Infrared Spectrophotometry <2.25>, and com-  
29 pare the spectrum with the Reference Spectrum or the spec-  
30 trum of Cabergoline RS: both spectra exhibit similar inten-  
31 sities of absorption at the same wave numbers. If any dif-  
32 ference appears between the spectra, dissolve Cabergoline  
33 and Cabergoline RS in ethanol (95), respectively, then  
34 evaporate the ethanol, dry the residues, and repeat the test  
35 on the residues.

36 **Optical rotation** <2.49>  $[\alpha]_D^{20}$ : -77 - -83° (0.1 g  
37 calculated on the anhydrous basis, ethanol (95), 50 mL, 100  
38 mm).

39 **Purity** (1) Heavy metals <1.07>—Proceed with 1.0 g of  
40 Cabergoline according to Method 4, and perform the test.

41 Prepare the control solution with 2.0 mL of Standard Lead  
42 Solution (not more than 20 ppm).

43 (2) Related substances—Conduct this procedure using  
44 light-resistant vessels. Perform the test with 20 μL of the  
45 sample solution obtained in the Assay as directed under  
46 Liquid Chromatography <2.01> according to the following  
47 conditions. Determine each peak area by the automatic in-  
48 tegration method, and calculate the amount of them by the  
49 area percentage method: the amounts of related substances  
50 A and B, having the relative retention times of about 0.8 and  
51 about 2.8 to cabergoline are not more than 0.5%, respec-  
52 tively, and the amount of the peak other than cabergoline  
53 and the peaks mentioned above is not more than 0.1%. Fur-  
54 thermore, the total amount of the peaks other than cabergo-  
55 line is not more than 1.5%.

56 **Operating conditions**—

57 Detector, column, column temperature, mobile phase,  
58 and flow rate: Proceed as directed in the operating  
59 conditions in the Assay.

60 Time span of measurement: About 4 times as long as the  
61 retention time of cabergoline, beginning after the solvent  
62 peak.

63 **System suitability**—

64 System performance: Proceed as directed in the system  
65 suitability in the Assay.

66 Test for required detectability: Use the diluted sample  
67 solution (1 in 500) as the solution for system suitability test.  
68 Pipet 5 mL of the solution for system suitability test, and  
69 add the mobile phase to make exactly 20 mL. Confirm that  
70 the peak area of cabergoline obtained with 20 μL of this  
71 solution is equivalent to 18 to 32% of that with 20 μL of the  
72 solution for system suitability test.

73 System repeatability: When the test is repeated 6 times  
74 with 20 μL of the solution for system suitability test under  
75 the above operating conditions, the relative standard  
76 deviation of the peak area of cabergoline is not more than  
77 2.0%.

78 **Water** <2.48> not more than 0.5% (1 g, volumetric titra-  
79 tion, direct titration).

80 **Assay** Conduct this procedure using light-resistant ves-  
81 sels. Weigh accurately about 30 mg each of Cabergoline  
82 and Cabergoline RS (separately determine the water <2.48>  
83 in the same manner as Cabergoline), dissolve each in the  
84 mobile phase to make exactly 25 mL, and use these solu-  
85 tions as the sample solution and the standard solution, re-  
86 spectively. Perform the test with exactly 20 μL each of the  
87 sample solution and standard solution as directed under Liq-  
88 uid Chromatography <2.01> according to the following con-  
89 ditions, and determine the peak areas, A<sub>T</sub> and A<sub>S</sub>, of caber-  
90 goline in each solution.

91 Amount (mg) of cabergoline ( $C_{26}H_{37}N_5O_2$ ) =  $M_S \times A_T /$   
 92  $A_S$

93  $M_S$ : Amount (mg) of Cabergoline RS taken, calculated on  
 94 the anhydrous basis

95 *Operating conditions*—

96 Detector: An ultraviolet absorption photometer  
 97 (wavelength: 280 nm).

98 Column: A stainless steel column 4.0 mm in inside  
 99 diameter and 25 cm in length, packed with  
 100 octadecylsilanized silica gel for liquid chromatography (10  
 101  $\mu\text{m}$  in particle diameter).

102 Column temperature: A constant temperature of about  
 103  $25^\circ\text{C}$ .

104 Mobile phase: Dissolve 6.8 g of potassium dihydrogen  
 105 phosphate in 900 mL of water, adjust to pH 2.0 with  
 106 phosphoric acid, and add water to make 1000 mL. To this  
 107 solution add 0.2 mL of triethylamine. To 840 mL of this  
 108 solution add 160 mL of acetonitrile.

109 Flow rate: Adjust so that the retention time of cabergoline  
 110 is about 12 minutes.

111 *System suitability*—

112 System performance: Suspend 50 mg of Cabergoline in  
 113 10 mL of 0.1 mol/L sodium hydroxide TS, and stir for 15  
 114 minutes. To 1 mL of this solution add 1 mL of 0.1 mol/L  
 115 hydrochloric acid TS, and add the mobile phase to make 10  
 116 mL. When the procedure is run with 20  $\mu\text{L}$  of this solution  
 117 under the above operating conditions, the resolution  
 118 between the peaks of related substance A having the relative  
 119 retention time of about 0.8 to cabergoline and cabergoline  
 120 is not less than 3.

121 System repeatability: When the test is repeated 6 times  
 122 with 20  $\mu\text{L}$  of the standard solution under the above  
 123 operating conditions, the relative standard deviation of the  
 124 peak area of cabergoline is not more than 1.0%.

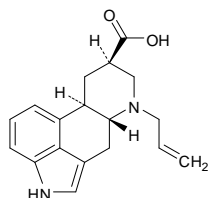
125 **Containers and storage** Containers—Tight containers.

126 Storage—Light-resistant.

127 **Others**

128 Related substance A:

129 (8*R*)-6-Allylergoline-8-carboxylic acid

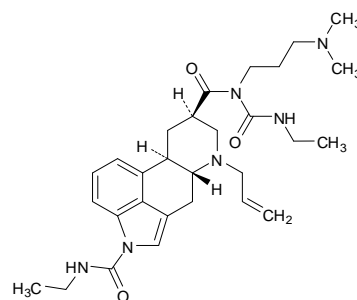


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131 Related substance B:

132 (8*R*)-6-Allyl-*N*-[3-(dimethylamino)propyl]-

133 *N*,1-bis(ethylcarbamoyl)ergoline-8-carboxamide



134

135 **Add the following to 9.01 Reference**

136 **Standards (1):**

137 **Cabergoline RS**

138