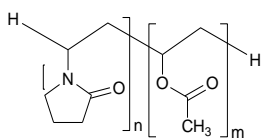


# 1 Copovidone

2 コポビドン



3  $(C_6H_9NO)_n(C_4H_6O_2)_m$

4 Poly[(2-oxopyrrolidin-1-yl)ethylene-co-(1-acetoxyethylene)]

5 [25086-89-9]

6

7 This monograph is harmonized with the European Pharmacopoeia and the U.S. Pharmacopoeia.

8 The corresponding part of the attributes/provisions which are agreed as non-harmonized within the scope of the harmonization is marked with symbols (◆ ◆), and the corresponding parts which are agreed as the JP local requirement other than the scope of the harmonization are marked with symbols (◇ ◇).

9 Copovidone is a copolymer of 1-vinyl-2-pyrrolidone and vinyl acetate at the ratio by weight of 3:2.

10 It contains not less than 35.3% and not more than 42.0% of vinyl acetate ( $C_4H_6O_2$ : 86.09), and not less than 7.0% and not more than 8.0% of nitrogen (N: 14.01), calculated on the dried basis.

11 The nominal K-value is shown on the label.

12 ◆**Description** Copovidone occurs as a white to yellowish white powder.

13 It is odorless or has a faint, characteristic odor.

14 It is very soluble in methanol and in ethanol (95), and freely soluble in water.

15 It is hygroscopic.◆

16 ◆**Identification** Determine the infrared absorption spectrum of Copovidone, previously dried, as directed in the potassium bromide disk method under Infrared Spectrophotometry <2.25>, and compare the spectrum with the Reference Spectrum: both spectra exhibit similar intensities of absorption at the same wave numbers.

17 **pH** <2.54> Dissolve 1.0 g of Copovidone in 10 mL of water: the pH of the solution is between 3.0 and 7.0.

18 **Purity** (1) Clarity and color of solution—Dissolve 1.0 g of Copovidone in 10 mL of water: the solution is colorless to pale yellow, or pale red, and it is clear or slightly opalescent.

19 ◇(2) Heavy metals <1.07>—Ignite 2.0 g of Copovidone as directed under Residue on Ignition Test <2.44>, add 2 mL of hydrochloric acid to the residue, then proceed accord-

20 ing to Method 2, and perform the test. Prepare the control solution with 2.0 mL of Standard Lead Solution (not more than 10 ppm).◇

21 (3) Aldehydes—Weigh accurately about 1 g of Copovidone, dissolve in 0.05 mol/L pyrophosphate buffer solution (pH 9.0) to make exactly 100 mL. Stopper, heat at 60°C for 60 minutes, allow to cool to room temperature, and use this solution as the sample solution. Separately, dissolve 0.140 g of acetaldehyde ammonia trimer trihydrate in water to make exactly 200 mL. Pipet 1 mL of this solution, add 0.05 mol/L pyrophosphate buffer solution (pH 9.0) to make exactly 100mL, and use this solution as the standard solution. Measure exactly 0.5 mL each of the sample solution, standard solution and water, transfer to separate 1-cm cells, add 2.5 mL of 0.05 mol/L pyrophosphate buffer solution (pH 9.0) and 0.2 mL of  $\beta$ -nicotinamide adenine dinucleotide TS to each of these cells, mix and stopper tightly. Allow to stand for 2 to 3 minutes at  $22 \pm 2^\circ\text{C}$ , and perform the test with these solutions as directed under Ultraviolet-visible Spectrophotometry <2.24> using water as the control solution. Determine the absorbances,  $A_{T1}$ ,  $A_{S1}$  and  $A_{B1}$ , of the subsequent solutions of the sample solution, the standard solution and water (blank) at 340 nm. Then, add 0.05 mL of aldehyde dehydrogenase TS to each of the cells, stir, and stopper tightly. Allow to stand at  $22 \pm 2^\circ\text{C}$  for 5 minutes. Determine the absorbances,  $A_{T2}$ ,  $A_{S2}$  and  $A_{B2}$ , of these solutions in the same manner as above: the content of aldehyde is not more than 500 ppm (as acetaldehyde).

22 Content (ppm) of aldehydes [as acetaldehyde ( $\text{CH}_3\text{CHO}$ )]  

$$= C/M \times \{ (A_{T2} - A_{T1}) - (A_{B2} - A_{B1}) \} / \{ (A_{S2} - A_{S1}) - (A_{B2} - A_{B1}) \} \times 100,000$$

23  $M$ : Amount (g) of Copovidone taken, calculated on the dried basis

24  $C$ : Concentration (mg/mL) of acetaldehyde in the standard solution, using 0.72 as conversion factor for acetaldehyde ammonia trimer trihydrate to acetaldehyde

25 (4) 1-Vinyl-2-pyrrolidone and vinyl acetate—Store the sample solution and standard solution at 5°C or below, and use within 8 hours. Weigh accurately about 0.25 g of Copovidone, dissolve in a mixture of water and acetonitrile (23:2) to make exactly 10 mL, and use this solution as the sample solution. Separately, dissolve 50 mg each of 1-vinyl-2-pyrrolidone and vinyl acetate in methanol to make exactly 100 mL. Pipet 1 mL of this solution and add methanol to make exactly 100 mL. Pipet 5 mL of this solution, add a mixture of water and acetonitrile (23:2) to make exactly 100 mL, and use this solution as the standard solution. Perform the test with exactly 20  $\mu\text{L}$  each of the sample solution and standard solution as directed under Liquid Chromatography <2.01> according to the following conditions, determine the peak areas,  $A_{Ta}$ ,  $A_{Tb}$ ,  $A_{Sa}$  and  $A_{Sb}$ , of 1-vinyl-

96 2-pyrrolidone and vinyl acetate in each solution, and calcu-  
 97 late the content of 1-vinyl-2-pyrrolidone and vinyl acetate  
 98 by the following equations: they are not more than 10 ppm.

$$99 \quad \text{Content (ppm) of 1-vinyl-2-pyrrolidone} \\
 100 \quad = A_{T_a}/A_{S_a} \times C_{S_a}/C_T \times 1000$$

$$101 \quad \text{Content (ppm) of vinyl acetate} \\
 102 \quad = A_{T_b}/A_{S_b} \times C_{S_b}/C_T \times 1000$$

103  $C_{S_a}$ : Concentration ( $\mu\text{g/mL}$ ) of 1-vinyl-2-pyrrolidone in  
 104 the standard solution

105  $C_{S_b}$ : Concentration ( $\mu\text{g/mL}$ ) of vinyl acetate in the stand-  
 106 ard solution

107  $C_T$ : Concentration (mg/mL) of Copovidone in the sample  
 108 solution, calculated on the dried basis

#### 109 *Operating conditions—*

110 Detector: An ultraviolet spectrophotometer (wavelength:  
 111 235 nm for 1-vinyl-2-pyrrolidone, 205 nm for vinyl  
 112 acetate).

113 Column: Two stainless steel columns, one is 4 mm in  
 114 inside diameter and 33 mm in length and the other is 4 mm  
 115 in inside diameter and 250 mm in length, packed with  
 116 octadecylsilanized silica gel for liquid chromatography (5  
 117  $\mu\text{m}$  in particle diameter), and use them as the pre-column  
 118 and the separation column, respectively.

119 Column temperature: A constant temperature of about  
 120  $40^\circ\text{C}$ .

121 Mobile phase: A mixture of water and acetonitrile (23:2).

122 Flow rate: 1.0 mL per minute (Retention times of 1-  
 123 vinyl-2-pyrrolidone and vinyl acetate are about 17 and  
 124 about 22 minutes, respectively).

125 Time span of measurement: For 40 minutes.

126 Washing of column: After each test with the sample  
 127 solution, elute and wash away remaining sample by passing  
 128 the mobile phase through the separation column or the pre-  
 129 column backwards at the flow rate mentioned above for 30  
 130 minutes.

#### 131 *System suitability—*

132 System performance: When the procedure is run with 20  
 133  $\mu\text{L}$  of the standard solution under the above operating  
 134 conditions (wavelength: 205 nm), 1-vinyl-2-pyrrolidone  
 135 and vinyl acetate are eluted in this order with the resolution  
 136 between these peaks being not less than 2.0.

137 System repeatability: When the test is repeated 6 times  
 138 with 20  $\mu\text{L}$  of the standard solution under the above  
 139 operating conditions, the relative standard deviations of the  
 140 peak areas of 1-vinyl-2-pyrrolidone and vinyl acetate are  
 141 not more than 2.0%, respectively.

142 (5) Peroxides — Weigh exactly an amount of Copo-  
 143 vidone, equivalent to 4.0 g calculated on the dried basis,  
 144 dissolve in water to make exactly 100 mL, and use this so-

145 lution as the sample solution. To 25 mL of the sample so-  
 146 lution add 2 mL of titanium (III) chloride-sulfuric acid TS,  
 147 and mix. Allow to stand for 30 minutes, and perform the  
 148 test with this solution as directed under Ultraviolet-visible  
 149 Spectrophotometry <2.24>, using a solution prepared by  
 150 adding 2 mL of diluted sulfuric acid (13 in 100) to 25 mL  
 151 of the sample solution as a blank: the absorbance of the  
 152 sample solution at 405 nm is not more than 0.35 (not more  
 153 than 400 ppm, as hydrogen peroxide).

154 (6) Hydrazine — Weigh exactly an amount of Copo-  
 155 vidone equivalent to 2.5 g calculated on the dried basis,  
 156 transfer to a 50-mL centrifuge tube, add 25 mL of water,  
 157 and stir to dissolve. Add 500  $\mu\text{L}$  of a solution of salicylal-  
 158 dehyde in methanol (1 in 20), stir and warm at  $60^\circ\text{C}$  for 15  
 159 minutes in a water bath. Allow to cool, add 2.0 mL of tolu-  
 160 ene, stopper tightly, shake vigorously for 2 minutes, centri-  
 161 fuge, and use the upper layer of the mixture as the sample  
 162 solution. Separately, dissolve 90 mg of salicylaldazine in  
 163 toluene to make exactly 100 mL. Pipet 1 mL of this solution,  
 164 add toluene to make exactly 100 mL, and use this solution  
 165 as the standard solution. Perform the test with these solu-  
 166 tions as directed under Thin-layer Chromatography <2.03>.  
 167 Spot 10  $\mu\text{L}$  each of the sample solution and standard solu-  
 168 tion on a plate of dimethylsilanized silica gel with fluores-  
 169 cent indicator for thin-layer chromatography. Develop the  
 170 plate with a mixture of methanol and water (2:1) to a dis-  
 171 tance of about three-fourths of the length of the plate, and  
 172 air-dry the plate. Examine under ultraviolet light (main  
 173 wavelength: 365 nm): the fluorescence of the spot obtained  
 174 from the sample solution corresponding to the spot having  
 175 a  $R_f$  value of about 0.3 from the standard solution is not  
 176 more intense than that of the spot from the standard solu-  
 177 tion (not more than 1 ppm).

178 (7) 2-Pyrrolidone — Weigh accurately about 1 g of Co-  
 179 povidone, add 5 mL of methanol for liquid chromatography,  
 180 and sonicate to dissolve. Add water to make exactly 100  
 181 mL, and use this solution as the sample solution. Separately,  
 182 dissolve 0.150 g of 2-pyrrolidone in a mixture of water and  
 183 methanol for liquid chromatography (19:1) to make exactly  
 184 100 mL. Pipet 3 mL of this solution, add a mixture of water  
 185 and methanol for liquid chromatography (19:1) to make ex-  
 186 actly 100 mL, and use this solution as the standard solution.  
 187 Perform the test with exactly 20  $\mu\text{L}$  each of the sample so-  
 188 lution and standard solution as directed under Liquid Chro-  
 189 matography <2.01> according to the following conditions,  
 190 and determine the peak areas,  $A_T$  and  $A_S$ , of 2-pyrrolidone  
 191 in each solution. Calculate the content of 2-pyrrolidone by  
 192 the following equation: not more than 0.5%.

$$193 \quad \text{Content (\%)} \text{ of 2-pyrrolidone} = A_T/A_S \times C_S/C_T \times \\
 194 \quad 100$$

195  $C_S$ : Concentration (mg/mL) of 2-pyrrolidone in the  
196 standard solution

197  $C_T$ : Concentration (mg/mL) of Copovidone in the sample  
198 solution, calculated on the dried basis

199 *Operating conditions*—

200 Detector: An ultraviolet absorption photometer  
201 (wavelength: 205 nm).

202 Column: Two stainless steel columns, one is 4.0 mm in  
203 inside diameter and 10 mm in length and the other is 4.6  
204 mm in inside diameter and 150 mm in length, packed with  
205 octadecylsilanized silica gel for liquid chromatography (5  
206  $\mu\text{m}$  in particle diameter), and use them as the pre-column  
207 and the separation column, respectively.

208 Column temperature: A constant temperature of about  
209  $40^\circ\text{C}$ .

210 Mobile phase: A mixture of water and methanol for  
211 liquid chromatography (19:1).

212 Flow rate: 0.8 mL per minute (retention time of 2-  
213 pyrrolidone is about 7 minutes).

214 Time span of measurement: For 30 minutes.

215 Washing of column: After each test with the sample  
216 solution, elute and wash away remaining sample by passing  
217 the mobile phase through the separation column or the pre-  
218 column backwards at the flow rate mentioned above for 30  
219 minutes.

220 *System suitability*—

221 System performance: When the procedure is run with 20  
222  $\mu\text{L}$  of the standard solution under the above operating  
223 conditions, the number of theoretical plates and the  
224 symmetry factor of the peak of 2-pyrrolidone are not less  
225 than 5000 and not more than 1.5, respectively.

226 System repeatability: When the test is repeated 6 times  
227 with 20  $\mu\text{L}$  of the standard solution under the above  
228 operating conditions, the relative standard deviation of the  
229 peak area of 2-pyrrolidone is not more than 2.0%.

230 **Loss on drying** <2.41> Not more than 5.0% (0.5 g,  $105^\circ\text{C}$ ,  
231 3 hours).

232 **Residue on ignition** <2.44> Not more than 0.1% (1 g).

233 **K-value** Weigh exactly an amount of Copovidone,  
234 equivalent to 1.000 g, calculated on the dried basis, dissolve  
235 in water to make exactly 100 mL, allow to stand for 60  
236 minutes, and use this solution as the sample solution. Per-  
237 form the test with the sample solution and with water at  
238  $25^\circ\text{C}$  as directed in Method 1 under the Viscosity Determin-  
239 ation <2.53>, and calculate the K-value by the following  
240 formula: the K-value of Copovidone is not less than 90.0%  
241 and not more than 110.0% of the nominal K-value.

$$242 \quad K = \frac{1.5 \log v_{\text{rel.}} - 1}{0.15 + 0.003c} +$$

$$243 \quad \frac{\sqrt{300c \log v_{\text{rel.}} + (c + 1.5c \log v_{\text{rel.}})^2}}{0.15c + 0.003c^2}$$

244  $c$ : Mass (g) of Copovidone in 100 mL of the solution,  
245 calculated on the dried basis

246  $v_{\text{rel.}}$ : Kinematic viscosity of the sample solution relative  
247 to that of water

248 **Assay (1)** Vinyl acetate—Weigh accurately about 2 g  
249 of Copovidone, add exactly 25 mL of 0.5 mol/L potassium  
250 hydroxide-ethanol TS and a few glass beads, heat under re-  
251 flux for 30 min. Titrate immediately with 0.5 mol/L hydro-  
252 chloric acid VS (indicator: 1 mL of phenolphthalein TS).  
253 Perform a blank determination in the same manner, and  
254 make any necessary correction.

$$255 \quad \text{Amount (\% of vinyl acetate)} \\ 256 \quad = 0.1 \times \frac{86.09}{56.11} \times \frac{28.05 (n_2 - n_1)}{M}$$

257  $M$ : Amount (g) of Copovidone taken, calculated on the  
258 dried basis

259  $n_1$ : Volume (mL) of 0.5 mol/L hydrochloric acid VS con-  
260 sumed in the blank test

261  $n_2$ : Volume (mL) of 0.5 mol/L hydrochloric acid VS con-  
262 sumed in the test

263 **(2)** Nitrogen—Weigh accurately about 0.1 g of Copo-  
264 vidone, and place in a Kjeldahl flask. Add 5 g of a pow-  
265 dered mixture of 33 g of potassium sulfate, 1 g of copper  
266 (II) sulfate pentahydrate and 1 g of titanium (IV) oxide,  
267 and wash down any adhering sample from the neck of the  
268 flask with a small amount of water. Add 7 mL of sulfuric  
269 acid allowing to flow down the inside wall of the flask.  
270 Heat the flask gradually until the solution has a clear, yel-  
271 low-green color, and the inside wall of the flask is free from  
272 a carbonized material, and then heat for further 45 minutes.  
273 After cooling, add cautiously 20 mL of water, and connect  
274 the flask to the distillation apparatus previously washed by  
275 passing steam through it. To the absorption flask add 30  
276 mL of a solution of boric acid (1 in 25), 3 drops of bromo-  
277 cresol green-methyl red TS and sufficient water to immerse  
278 the lower end of the condenser tube. Add 30 mL of a solu-  
279 tion of sodium hydroxide (2 in 5) through the funnel, rinse  
280 cautiously the funnel with 10 mL of water, immediately  
281 close the clamp attached to the rubber tube, then start the  
282 distillation with steam to obtain 80 to 100 mL of the distil-  
283 late. Remove the absorption flask from the lower end of the  
284 condenser tube, rinsing the end part with a small quantity  
285 of water, and titrate <2.50> the distillate with 0.025 mol/L  
286 sulfuric acid VS until the color of the solution changes from  
287 green through pale grayish blue to pale grayish red-purple.  
288 Perform a blank determination in the same manner, and  
289 make any necessary correction.

290 Each mL of 0.025 mol/L sulfuric acid VS = 0.700 mg of N

291 ♦ **Containers and storage** Containers—Tight containers.

292 ♦

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