## 1 Triazolam

2 トリアゾラム



4  $C_{17}H_{12}Cl_2N_4$ : 343.21

- $5 \quad \ \ 8-Chloro-6-(2-chlorophenyl)-1-methyl-4H-[1,2,4]triazolo[4,3-a][1,4]$
- 6 benzodiazepine
- 7 [28911-01-5]

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9 Triazolam, when dried, contains not less than 10 98.0% and not more than 102.0% of triazolam 11  $(C_{17}H_{12}Cl_2N_4)$ .

12 Description Triazolam occurs as a white crystalline13 powder.

14 It is sparingly soluble in N,N-dimethylformamide,

- 15 slightly soluble in ethanol (95), and practically insoluble in16 water.
- 17 It shows crystal polymorphism.

18 Identification (1) Determine the absorption spectrum 19 of a solution of Triazolam in ethanol (95) (1 in 200,000) as directed under Ultraviolet-visible Spectrophotometry 20 <2.24>, and compare the spectrum with the Reference Spec-21 22 trum or the spectrum of a solution of Triazolam RS pre-23 pared in the same manner as the sample solution: both spec-24 tra exhibit similar intensities of absorption at the same 25 wavelengths. 26 (2) Determine the infrared absorption spectrum of Tri-27 azolam, previously dried, as directed in the paste method under Infrared Spectrophotometry <2.25>, and compare the 28 29 spectrum with the Reference Spectrum or the spectrum of

dried Triazolam RS: both spectra exhibit similar intensitiesof absorption at the same wave numbers.

32 (3) Perform the test with Triazolam as directed under
33 Flame Coloration Test <1.04> (2): a green to green-blue
34 color appears.

35 **Melting point** <2.60> 239 – 243°C.

36 Purity (1) Chloride <1.03> — To 1.0 g of Triazolam add
37 50 mL of water, and allow to stand for 1 hour while occa38 sional shaking, and filter. Discard the first 10 mL of the
39 filtrate, pipet 25 mL of the subsequent filtrate, and add 6

40 mL of dilute nitric acid and water to make 50 mL. Perform

41 the test using this solution as the test solution. Prepare the

42 control solution with 0.40 mL of 0.01 mol/L hydrochloric

43 acid VS (not more than 0.028%).

44 (2) Heavy metals – Being specified separately when45 the drug is granted approval based on the Law.

46 (3) Related substances – Dissolve 0.14 g of Triazolam in 10 mL of N.N-dimethylformamide, and use this solution 47 as the sample solution. Pipet 1 mL of the sample solution, 48 49 add N,N-dimethylformamide to make exactly 100 mL, and 50 use this solution as the standard solution. Perform the test with exactly 12  $\mu$ L each of the sample solution and stand-51 52 ard solution as directed under Liquid Chromatography 53 <2.01> according to the following conditions. Determine 54 each peak area by the automatic integration method: the 55 area of peak other than triazolam obtained from the sample 56 solution is not larger than 1/5 times the peak area of tria-57 zolam from the standard solution, and the total area of the 58 peaks other than triazolam is not larger than the peak area 59 of triazolam from the standard solution. For the areas of the 60 peaks, related substance A having the retention time of about 0.7 to triazolam, related substance B having the re-61 62 tention time of about 1.5, and related substance C having 63 the retention time of about 2.4, multiply their relative re-64 sponse factors, 1.8, 0.6 and 4.3, respectively. 65 Operating conditions -

66 Detector, column, column temperature, mobile phase A,

- mobile phase B, flowing of mobile phase, and flow rate:Proceed as directed in the operating conditions in the Assay.
- 69 Time span of measurement: For 39 minutes after
- 70 injection, beginning after the solvent peak.
- 71 System suitability –

Test for required detectability: Pipet 1 mL of the standard solution, and add *N*,*N*-dimethylformamide to make exactly 10 mL. Confirm that the peak area of triazolam obtained with 12  $\mu$ L of this solution is equivalent to 7 to 13% of that with 12  $\mu$ L of the standard solution.

77 System performance: When the procedure is run with 12 78  $\mu$ L of the standard solution under the above operating 79 conditions, the number of theoretical plates and the 80 symmetry factor of the peak of triazolam are not less than 81 4500 and not more than 1.6, respectively.

82 System repeatability: When the test is repeated 6 times 83 with 12  $\mu$ L of the standard solution under the above 84 operating conditions, the relative standard deviation of the 85 peak area of triazolam is not more than 2.0%.

by peak area of thazolam is not more than 2.0%.

86 Loss on drying <2.41> Not more than 0.5% (1 g, 105°C,
87 4 hours).

88 **Residue on ignition**  $\langle 2.44 \rangle$  Not more than 0.3% (1 g).

89 Assay Weigh accurately about 55 mg each of Triazolam

- 90 and Triazolam RS, previously dried, dissolve each in *N*,*N*-
- 91 dimethylformamide to make exactly 50 mL, and use these

92 solutions as the sample solution and the standard solution,

93 respectively. Perform the test with 12  $\mu$ L each of the sam-

94 ple solution and standard solution as directed under Liquid

95 Chromatography <2.01> according to the following condi-

96 tions, and determine the peak areas,  $A_{\rm T}$  and  $A_{\rm S}$ , of triazolam 97 in each solution.

97 III each solu

98 Amount (mg) of triazolam (C<sub>17</sub>H<sub>12</sub>Cl<sub>2</sub>N<sub>4</sub>)  
99 
$$=M_{\rm S} \times A_{\rm T} / A_{\rm S}$$

100  $M_{\rm S}$ : Amount (mg) of Triazolam RS taken

## 101 Operating conditions -

102	Detector:	An	ultraviolet	absorption	photometer
103	(wavelength:	254 n	ım).		

- 104Column: A stainless steel column 4.6 mm in inside105diameter and 25 cm in length, packed with phenylsilanized106silica gel for liquid chromatography (5  $\mu$ m in particle
- 107 diameter).
  108 Column temperature: A constant temperature of about
  109 40°C.
- 110 Mobile phase A: A mixture of methanol and diluted 111 acetic acid-ammonium acetate buffer solution (pH 4.5) (1 112 in 10) (14:11).
- 113 Mobile phase B: A mixture of methanol and diluted 114 acetic acid-ammonium acetate buffer solution (pH 4.5) (1
- 115 in 10) (19:1).

116 Flowing of mobile phase: Control the gradient by mixing

117 the mobile phases A and B as directed in the following table.

Time after injection of sample (min)	Mobile phase A (vol%)	Mobile phase B (vol%)
0 - 14	98	2
14 - 34	$98 \rightarrow 1$	$2 \rightarrow 99$
34 - 39	1	99

120 Flow rate: 2.0 mL per minute.

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121 System suitability -
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122 System performance: When the procedure is run with 12 123  $\mu$ L of the standard solution under the above operating 124 conditions, the number of theoretical plates and the 125 symmetry factor of the peak of triazolam are not less than 126 4500 and not more than 2.0, respectively.

127 System repeatability: When the test is repeated 6 times 128 with 12  $\mu$ L of the standard solution under the above 129 operating conditions, the relative standard deviation of the 130 peak area of triazolam is not more than 1.0%.

## 131 Containers and storage Containers – Tight containers.

## 132 Others

- 133 Related substance A:
- 134 3-Amino-6-chloro-4-(2-chlorophenyl)-2-methyl-3,4-dihy-
- 135 droquinazolin-4-ol



- 137 Related substance B:
- 138 8-Chloro-6-(2-chlorophenyl)-1-ethenyl-4H-[1,2,4]tria-
- 139 zolo[4,3-*a*][1,4]benzodiazepine



- 141 Related substance C:
- 142 8-Chloro-6-(2-chlorophenyl)-6-methoxy-1-methyl-
- 143 4*H*,6*H*-[1,2,4]triazolo[4,3-*a*][4,1]benzoxazepine



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146 Add the following to 9.01 Reference

147 Standards (1):

148 Triazolam RS