

1 2.05 Size-Exclusion Chromatography

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4 Size-exclusion chromatography is a separation technique
5 of liquid chromatography, which separate molecules in so-
6 lution based on their size. It is used for the determination of
7 the molecular mass of high-molecular mass compounds
8 such as polysaccharides, nucleic acids, proteins and chemi-
9 cally synthesized polymers, determination of the molecular
10 mass distribution, purity tests and so on. A method using an
11 aqueous solvent as the mobile phase for water-soluble high-
12 molecular compounds is also called gel filtration chroma-
13 tography. A method using an organic solvent as the mobile
14 phase is also called gel permeation chromatography. Here
15 methods using an aqueous solvent as the mobile phase are
16 described. The principle of the separation is the same when
17 using an organic solvent as the mobile phase.

18 1. Principle of separation

19 Components to be tested are separated according to the
20 easiness of entry into the pores of the packing material of
21 the column. Molecules larger than the largest pore move
22 rapidly through the spaces between the particles of the
23 packing material without entering the pores, and elute at the
24 position of the retention volume of an unretained compound
25 (V_0) of the column. Molecules smaller than the pores enter
26 into the pores depending on their sizes, and smaller mole-
27 cules elute later because they enter deeper into the pores.
28 Small molecules that can freely enter all pores elute last at
29 the position of a completely permeated compound, or the
30 total mobile phase volume (V_t). The elution position of a
31 molecule will be influenced not only by molecular mass but
32 also by the structure of the molecule, solvent and interaction
33 between the molecule and the packing material, etc.

34 2. Apparatus

35 A liquid chromatograph is usually used. A porous pack-
36 ing material is used for a column. As packing materials, sil-
37 ica particles whose surface are coated with hydrophilic
38 modification, or cross-linked hydrophilic polymers are used.
39 Since measurable molecular mass range differs depending
40 on the pore size of a packing material and its distribution,
41 select an appropriate column. In order to expand the meas-
42 urable molecular mass range, a column may be connected
43 with another column having the different target molecular
44 mass range. As a mobile phase, a buffer solution etc. is used.
45 It is important to select appropriately a mobile phase to sup-
46 press interaction other than the principle of size exclusion
47 between a packing material and a component to be tested.
48 Adjustment of pH, addition of salt, etc. may be useful to
49 suppress electrostatic interaction between a packing mate-
50 rial and a component to be tested, and addition of an organic

51 solvent (methanol, acetonitrile, etc.) may be useful to sup-
52 press hydrophobic interaction. The flow rate of a mobile
53 phase, column temperature, sample injection volume, and
54 the concentration of a sample solution should be set appro-
55 priately because they affect the separation. As a detector, an
56 ultraviolet-visible spectrophotometer, a differential refrac-
57 tometer, a static light scattering detector, a dynamic light
58 scattering detector, an evaporative light scattering detector,
59 etc. are used. Usually, the molecular mass of a component
60 to be tested is determined by comparing the elution position
61 with molecular mass standards, but when using a static light
62 scattering detector, the molecular mass of molecules in an
63 eluate can be obtained directly.

64 3. Procedure

65 Follow Liquid Chromatography <2.01>.

66 4. Measurement of molecular mass

67 When determining molecular mass, average molecular
68 mass or molecular mass distribution by size exclusion chro-
69 matography, measure a sample solution and molecular mass
70 standard solutions prepared using appropriate molecular
71 mass standards under the same test conditions, and deter-
72 mine as follows, unless otherwise specified. The molecular
73 mass standards should have the same physical properties as
74 a component to be tested. The obtained molecular mass
75 value depends on the molecular mass standards used and
76 analysis conditions, and is a relative value to the molecular
77 mass standards.

78 4.1. Molecular mass of a monodisperse component

79 A molecular mass calibration curve is prepared by plot-
80 ting the relationship of retention volume (or retention time)
81 and the logarithmic value of molecular mass labelled on
82 molecular mass standards. The molecular mass will be de-
83 termined from the molecular mass calibration curve for the
84 retention volume (or retention time) obtained from the chro-
85 matogram of a sample solution. Usually, the molecular
86 mass of a component to be tested should be within the range
87 of the molecular mass calibration curve.

88 4.2. Average molecular mass of a polydisperse compo- 89 nent

90 A molecular mass calibration curve is prepared from
91 chromatograms obtained from molecular mass standard so-
92 lutions. A chromatogram obtained from a sample solution
93 is divided, and the molecular mass of each eluted fraction is
94 obtained from the molecular mass calibration curve. The
95 concentration or amount of the component in each fraction
96 is determined, and the number-average molecular mass
97 (M_n), mass-average molecular mass (M_w) and dispersity (d)
98 of the sample are calculated by the following equations.

99 The dispersity may be an indicator of the width of molec-
100 ular mass distribution.

101

102
$$M_n = \frac{\sum M_i N_i}{\sum N_i} = \frac{\sum C_i}{\sum \frac{C_i}{M_i}} = \frac{1}{\sum \frac{w_i}{M_i}}$$

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104
$$M_w = \frac{\sum M_i^2 N_i}{\sum M_i N_i} = \frac{\sum C_i M_i}{\sum C_i} = \sum w_i M_i,$$

105

$$d = \frac{M_w}{M_n}$$

106

107 M_i : Molecular mass of i-th fraction

108 C_i : Concentration of i-th fraction

109 N_i : Number of molecules in i-th fraction

110 w_i : Mass fraction of i-th fraction ($w_i = \frac{M_i N_i}{\sum M_i N_i} = \frac{C_i}{\sum C_i}$)

111 4.3. Molecular mass distribution

112 As a distribution curve showing molecular mass distribu-
 113 tion, an integral molecular mass distribution curve plotting
 114 the logarithmic value of molecular mass on the abscissa and
 115 the integral value of mass fraction on the ordinate, and a
 116 differential molecular mass distribution curve plotting the
 117 logarithmic value of molecular mass on the abscissa and the
 118 slope of the integral molecular mass distribution curve de-
 119 termined at each molecular mass on the ordinate, are used.

120 Specifications for molecular mass distribution can be
 121 shown in a form depending on the purpose, such as mass
 122 average molecular mass, dispersity, mass fractions of mol-
 123 ecules in a specific molecular mass range.

124 5. Points to consider on system suitability and changes 125 in operating conditions

126 Liquid Chromatography <2.01> is applied to points to
 127 consider on specifications of system suitability and changes
 128 in operating conditions.