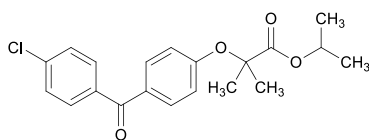


## 1 Fenofibrate

2 フェノフィブラート



3

4  $C_{20}H_{21}ClO_4$  : 360.83

5 1-Methylethyl 2-[4-(4-chlorobenzoyl)phenoxy]-2-methylpropanoate

6 [49562-28-9]

7

8 Fenofibrate, when dried, contains not less than  
9 98.5% and not more than 101.0% of fenofibrate  
10 ( $C_{20}H_{21}ClO_4$ ).

11 **Description** Fenofibrate occurs as a white to pale yellow-  
12 white crystalline powder.

13 It is soluble in ethanol (99.5), and practically insoluble in  
14 water.

15 It shows crystal polymorphism.

16 **Identification** (1) Determine the absorption spectrum  
17 of a solution of Fenofibrate in ethanol (99.5) (1 in 80,000)  
18 as directed under Ultraviolet-visible Spectrophotometry  
19 <2.24>, and compare the spectrum with the Reference Spec-  
20 trum or the spectrum of a solution of Fenofibrate RS pre-  
21 pared in the same manner as the sample solution: both spec-  
22 tra exhibit similar intensities of absorption at the same  
23 wavelengths.

24 (2) Determine the infrared absorption spectrum of Fen-  
25 ofibrate, previously dried, as directed in the potassium bro-  
26 mide disk method under Infrared Spectrophotometry <2.25>,  
27 and compare the spectrum with the Reference Spectrum or  
28 the spectrum of dried Fenofibrate RS: both spectra exhibit  
29 similar intensities of absorption at the same wave numbers.

30 (3) Perform the test with Fenofibrate as directed under  
31 Flame Coloration Test <1.04>: a green color appears.

32 **Melting point** <2.60> 80 – 83°C

33 **Purity** (1) Heavy metals <1.07>—Proceed with 1.0 g of  
34 Fenofibrate according to Method 4, and perform the test.  
35 Prepare the control solution with 2.0 mL of Standard Lead  
36 Solution (not more than 20 ppm).

37 (2) Related substances—Conduct this procedure using  
38 light-resistant vessels. Dissolve 0.10 g of Fenofibrate in a  
39 mixture of acetonitrile and water (7:3) to make 100 mL. To  
40 5 mL of this solution add a mixture of acetonitrile and water  
41 (7:3) to make 25 mL, and use this solution as the sample  
42 solution. Pipet 3 mL of the sample solution, add a mixture  
43 of acetonitrile and water (7:3) to make exactly 50 mL. Pipet  
44 2.5 mL of this solution, add a mixture of acetonitrile and  
45 water (7:3) to make exactly 50 mL, and use this solution as

46 the standard solution. Perform the test with exactly 20  $\mu$ L  
47 each of the sample solution and standard solution as di-  
48 rected under Liquid Chromatography <2.01> according to  
49 the following conditions, and determine each peak area by  
50 the automatic integration method: the peak area of the re-  
51 lated substance A having the relative retention time of about  
52 1.4 to fenofibrate obtained from the sample solution is not  
53 larger than 4/5 times the peak area of fenofibrate from the  
54 standard solution, and the total area of the peaks other than  
55 fenofibrate from the sample solution is not larger than the  
56 peak area of fenofibrate from the standard solution.

57 **Operating conditions**—

58 Detector, column, column temperature, mobile phase,  
59 and flow rate: Proceed as directed in the operating  
60 conditions in the Assay.

61 Time span of measurement: About 2 times as long as the  
62 retention time of fenofibrate, beginning after the solvent  
63 peak.

64 **System suitability**—

65 Test for required detectability: Pipet 5 mL of the standard  
66 solution, and add a mixture of acetonitrile and water (7:3)  
67 to make exactly 25 mL. Confirm that the peak area of feno-  
68 fibrate obtained with 20  $\mu$ L of this solution is equivalent to  
69 15 to 25% of that with 20  $\mu$ L of the standard solution.

70 System performance: Dissolve 0.10 g each of Fenofibrate  
71 and 4-chlorobenzophenon in 100 mL of a mixture of ace-  
72 tonitrile and water (7:3). To 2 mL of this solution add a mix-  
73 ture of acetonitrile and water (7:3) to make 50 mL. To 1 mL  
74 of this solution add a mixture of acetonitrile and water (7:3)  
75 to make 50 mL. When the procedure is run with 20  $\mu$ L of  
76 this solution under the above operating conditions, 4-chlo-  
77 robenzophenon and fenofibrate are eluted in this order with  
78 the resolution between these peaks being not less than 10.

79 System repeatability: When the test is repeated 6 times  
80 with 20  $\mu$ L of the standard solution under the above operat-  
81 ing conditions, the relative standard deviation of the peak  
82 area of fenofibrate is not more than 5%.

83 **Loss on drying** <2.41> Not more than 0.5% (1 g, in vac-  
84 uum, phosphorus (V) oxide, 60°C, 4 hours).

85 **Residue on ignition** <2.44> Not more than 0.1% (1 g).

86 **Assay** Conduct this procedure using light-resistant ves-  
87 sels. Weigh accurately about 50 mg each of Fenofibrate and  
88 Fenofibrate RS, both previously dried, dissolve each in a  
89 mixture of acetonitrile and water (7:3) to make exactly 50  
90 mL. Pipet 2 mL each of these solutions, add exactly 2 mL  
91 of the internal standard solution to each, add a mixture of  
92 acetonitrile and water (7:3) to make 50 mL, and use these  
93 solutions as the sample solution and the standard solution,  
94 respectively. Perform the test with 20  $\mu$ L each of the sample  
95 solution and standard solution as directed under Liquid

96 Chromatography <2.01> according to the following condi-  
 97 tions, and calculate the ratios,  $Q_T$  and  $Q_S$ , of the peak area  
 98 of fenofibrate to the internal standard.

$$\begin{aligned} 99 & \text{Amount (mg) of fenofibrate (C}_{20}\text{H}_{21}\text{ClO}_4\text{)} \\ 100 & = M_S \times Q_T / Q_S \end{aligned}$$

101  $M_S$ : Amount (mg) of Fenofibrate RS taken

102 *Internal standard solution*—A solution of 4-chlorobenzophen-  
 103 on in a mixture of acetonitrile and water (7:3) (11 in  
 104 10,000).

105 *Operating conditions*—

106 *Detector*: An ultraviolet absorption photometer  
 107 (wavelength: 286 nm).

108 *Column*: A stainless steel column 4.6 mm in inside  
 109 diameter and 15 cm in length, packed with  
 110 octadecylsilanized silica gel for liquid chromatography (5  
 111  $\mu\text{m}$  in particle diameter).

112 *Column temperature*: A constant temperature of about  
 113 40°C.

114 *Mobile phase*: A mixture of acetonitrile and 0.02 mol/L  
 115 phosphate buffer solution (pH 3.0) (7:3).

116 *Flow rate*: Adjust so that the retention time of fenofibrate  
 117 is about 8 minutes.

118 *System suitability*—

119 *System performance*: When the procedure is run with 20  
 120  $\mu\text{L}$  of the standard solution under the above operating  
 121 conditions, the internal standard and fenofibrate are eluted  
 122 in this order with the resolution between these peaks being  
 123 not less than 10.

124 *System repeatability*: When the test is repeated 6 times  
 125 with 20  $\mu\text{L}$  of the standard solution under the above  
 126 operating conditions, the relative standard deviation of the  
 127 ratio of the peak area of fenofibrate to that of the internal  
 128 standard is not more than 1.0%.

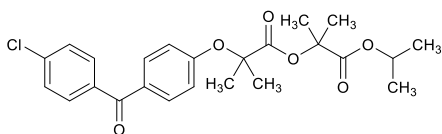
129 **Containers and storage** Containers—Tight containers.

130 Storage—Light-resistant.

### 131 Others

132 Related substance A:

133 2-Methyl-1-(1-methylethoxy)-1-oxopropan-2-yl 2-[4-(4-  
 134 chlorobenzoyl)phenoxy]-2-methylpropanoate



135

136

137 **Add the following to 9.01 Reference**

138 **Standards (1):**

139 **Fenofibrate RS**

140 **Add the following to 9.41 Reagents,**

141 **Test Solutions:**

142 **4-Chlorobenzophenon**  $\text{C}_6\text{H}_5\text{COC}_6\text{H}_4\text{Cl}$  A white,  
 143 crystalline powder or powder.

144 *Identification*—Determine the absorption spectrum of a  
 145 solution of 4-chlorobenzophenon in ethanol (99.5) (3 in  
 146 50,000) as directed under Ultraviolet-visible Spectropho-  
 147 tometry <2.24>: it exhibits a maximum between 256 nm and  
 148 260 nm.

149 *Melting point*: 73 – 78°C

150 *Content*: not less than 98.0%. *Assay*—Dissolve 1 g of  
 151 4-chlorobenzophenon in acetone to make 10 mL, and use  
 152 this solution as the sample solution. Perform the test with 1  
 153  $\mu\text{L}$  of the sample solution as directed under Gas Chroma-  
 154 tography <2.02> according to the following conditions. De-  
 155 termine each peak area by the automatic integration method,  
 156 and calculate the amount of 4-chlorobenzophenon by the  
 157 area percentage method.

158 *Operating conditions*

159 *Detector*: A hydrogen flame-ionization detector.

160 *Column*: A fused silica column 0.25 mm in inside  
 161 diameter and 30 m in length, coated with  
 162 dimethylpolysiloxane for gas chromatography in thickness  
 163 of 0.25  $\mu\text{m}$ .

164 *Column temperature*: A constant temperature of about  
 165 220°C.

166 *Injection port temperature*: 270°C.

167 *Detector temperature*: 250°C.

168 *Carrier gas*: Helium.

169 *Flow rate*: 1.33 mL per minute.

170 *Split ratio*: 1:100.

171 *Time span of measurement*: About 3 times as long as the  
 172 retention time of 4-chlorobenzophenon.

173 *System suitability*

174 *System performance*: To 1 mL of the sample solution add  
 175 acetone to make 10 mL. When the procedure is run with 1  
 176  $\mu\text{L}$  of this solution under the above operating conditions,  
 177 the number of theoretical plates and the symmetry factor of  
 178 the peak of 4-chlorobenzophenon are not less than 50,000  
 179 and not more than 1.2, respectively.

180 *System repeatability*: To 1 mL of the sample solution add  
 181 acetone to make 10 mL. When the test is repeated 6 times  
 182 with 1  $\mu\text{L}$  of this solution under the above operating  
 183 conditions, the relative standard deviation of the peak area  
 184 of 4-chlorobenzophenon is not more than 2.0%.