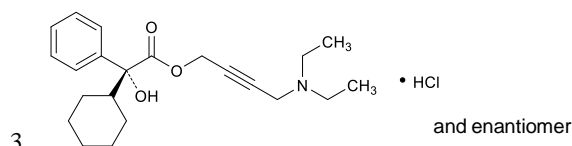


# 1 Oxybutynin Hydrochloride

2 オキシブチニン塩酸塩



4  $C_{22}H_{31}NO_3 \cdot HCl$ : 393.95

5 4-(Diethylamino)but-2-yn-1-yl (2*RS*)-2-cyclohexyl-2-hydroxy-  
6 2-phenylacetate monohydrochloride

7 [1508-65-2]

8

9 Oxybutynin Hydrochloride, when dried, contains not  
10 less than 98.0% and not more than 101.0% of oxy-  
11 butynin hydrochloride ( $C_{22}H_{31}NO_3 \cdot HCl$ ).

12 **Description** Oxybutynin Hydrochloride occurs as a white  
13 crystalline powder.

14 It is freely soluble in water and in ethanol (99.5).

15 A solution of Oxybutynin Hydrochloride (1 in 50) shows  
16 no optical rotation.

17 **Identification (1)** Determine the absorption spectrum of  
18 a solution of Oxybutynin Hydrochloride (3 in 100,000) as di-  
19 rected under Ultraviolet-visible Spectrophotometry <2.24>,  
20 and compare the spectrum with the Reference Spectrum: both  
21 spectra exhibit similar intensities of absorption at the same  
22 wavelengths.

23 **(2)** Determine the infrared absorption spectrum of Oxy-  
24 butynin Hydrochloride, previously dried, as directed in the  
25 potassium chloride disk method under Infrared Spectropho-  
26 tometry <2.25>, and compare the spectrum with the Refer-  
27 ence Spectrum: both spectra exhibit similar intensities of ab-  
28 sorption at the same wave numbers.

29 **(3)** A solution of Oxybutynin Hydrochloride (1 in 50) re-  
30 sponds to Qualitative Tests <1.09> for chloride.

31 **Melting point** <2.60> 124 – 129°C

32 **Purity (1)** Heavy metals <1.07>—Proceed with 2.0 g of  
33 Oxybutynin Hydrochloride according to Method 1, and per-  
34 form the test. Prepare the control solution with 3.0 mL of  
35 Standard Lead Solution (not more than 15 ppm).

36 **(2)** Related substances—Dissolve 50 mg of Oxybutynin  
37 Hydrochloride in 10 mL of the mobile phase, and use this  
38 solution as the sample solution. Pipet 1 mL of the sample so-  
39 lution, add the mobile phase to make exactly 200 mL, and use  
40 this solution as the standard solution. Perform the test with  
41 exactly 10  $\mu$ L each of the sample solution and standard solu-  
42 tion as directed under Liquid Chromatography <2.01> ac-  
43 cording to the following conditions, and determine each peak  
44 area by the automatic integration method: the peak area of the

45 related substance A, having the relative retention time of  
46 about 1.6 to oxybutynin, obtained from the sample solution  
47 is not larger than 3 times the peak area of oxybutynin from  
48 the standard solution, and the area of the peak other than ox-  
49 ybutynin and the peak mentioned above from the sample so-  
50 lution is not larger than 1/5 times the peak area of oxybutynin  
51 from the standard solution. Furthermore, the total area of the  
52 peaks other than oxybutynin and the related substance A from  
53 the sample solution is not larger than the peak area of oxy-  
54 butynin from the standard solution. For the peak area of the  
55 related substance A, multiply the relative response factor, 2.3.

56 **Operating conditions**—

57 **Detector:** An ultraviolet absorption photometer  
58 (wavelength: 210 nm).

59 **Column:** A stainless steel column 3.9 mm in inside  
60 diameter and 15 cm in length, packed with octylsilanized  
61 silica gel for liquid chromatography (5  $\mu$ m in particle  
62 diameter).

63 **Column temperature:** A constant temperature of about  
64 25°C.

65 **Mobile phase:** Dissolve 3.4 g of potassium dihydrogen  
66 phosphate and 4.36 g of dipotassium hydrogen phosphate in  
67 water to make 1000 mL. To 490 mL of this solution add 510  
68 mL of acetonitrile for liquid chromatography.

69 **Flow rate:** Adjust so that the retention time of oxybutynin  
70 is about 15 minutes.

71 **Time span of measurement:** About 2 times as long as the  
72 retention time of oxybutynin.

73 **System suitability**—

74 **Test for required detectability:** Pipet 2 mL of the standard  
75 solution, and add the mobile phase to make exactly 20 mL.  
76 Confirm that the peak area of oxybutynin obtained with 10  
77  $\mu$ L of this solution is equivalent to 7 to 13% of that with 10  
78  $\mu$ L of the standard solution.

79 **System performance:** When the procedure is run with 10  
80  $\mu$ L of the standard solution under the above operating condi-  
81 tions, the number of theoretical plates and the symmetry fac-  
82 tor of the peak of oxybutynin are not less than 5000 and not  
83 more than 1.5, respectively.

84 **System repeatability:** When the test is repeated 6 times  
85 with 10  $\mu$ L of the standard solution under the above operating  
86 conditions, the relative standard deviation of the peak area of  
87 oxybutynin is not more than 2.0%.

88 **Loss on drying** <2.41> Not more than 3.0% (0.5 g, 105°C,  
89 4 hours).

90 **Residue on ignition** <2.44> Not more than 0.1% (1 g).

91 **Assay** Weigh accurately about 0.5 g of Oxybutynin Hydro-  
92 chloride, previously dried, dissolve in 70 mL of a mixture of  
93 acetic anhydride and acetic acid (100) (7:3), and titrate

94 <2.50> with 0.1 mol/L perchloric acid VS (potentiometric ti-  
95 tration). Perform a blank determination in the same manner,  
96 and make any necessary correction.

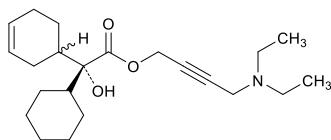
97 Each mL of 0.1 mol/L perchloric acid VS  
98 = 39.40 mg of C<sub>22</sub>H<sub>31</sub>NO<sub>3</sub>.HCl

99 **Containers and storage** Containers—Tight containers.  
100 Storage—Light-resistant.

101 **Others**

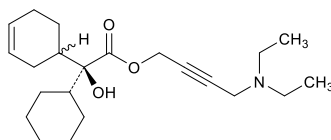
102 Related substance A:

103 4-(Diethylamino)but-2-yn-1-yl (2*R*)-2-(cyclohex-3-en-1-yl)-  
104 2-cyclohexyl-2-hydroxyacetate



105

106 4-(Diethylamino)but-2-yn-1-yl (2*S*)-2-(cyclohex-3-en-1-yl)-  
107 2-cyclohexyl-2-hydroxyacetate



108