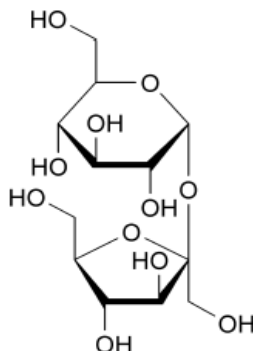


1
2
3
4
5
6
7
8
9

Rev. 1, Stage 2

SUCROSE

10
11
12
13
14
15
16
17
18
19 $C_{12}H_{22}O_{11}$

Mr 342.30

 β -D-Fructofuranosyl α -D-glucopyranoside [57-50-1]

DEFINITION

Sucrose is a sugar obtained from *Saccharum officinarum* Linné (Fam. Gramineae), *Beta vulgaris* Linné (Fam. Chenopodiaceae), and other sources. It contains no added substances.

24

IDENTIFICATION

A. Infrared absorption spectrophotometry – KBr or ATR

Record the infrared absorption spectrum of Sucrose and compare with the reference spectrum or the spectrum obtained with the reference standard: the transmission minima or absorption maxima correspond in position and relative size.

B. Examine the chromatograms obtained in the assay.

The principal peak in the chromatogram obtained with the test solution is similar in

31

32 retention time and size to the principal peak in the chromatogram obtained with Sucrose
33 standard solution.

34

35 **Solution S.** Dissolve 50.0 g in *water* and dilute to 100 ml with the same
36 solvent.

37

38 **Appearance of solution.** Solution S is clear.

39 Its clarity is the same as that of water or its opalescence is not more pronounced
40 than that of reference suspension I.

41

42 **Conductivity:** maximum $35 \mu\text{S}\cdot\text{cm}^{-1}$ at 20°C .

43 Dissolve 31.3 g in *carbon dioxide-free water* prepared from *distilled water* and dilute
44 to 100 ml with the same solvent. Measure the conductivity of the solution (C_1),
45 while gently stirring with a magnetic stirrer, and that of the water used for
46 preparing the solution (C_2). The readings must be stable within 1 per cent over a
47 period of 30 s. Calculate the conductivity of the solution of the substance to be
48 examined from the expression: $C_1 - 0.35 C_2$

49

50 **Specific optical rotation:** +66.3 to +67.0.

51 Dissolve 26.0 g in *water* and dilute to 100.0 ml with the same solvent.

52

53

54 **Related substances**

55 *Diluent:* Water

56 *System suitability solution:* 10 mg/mL of Sucrose reference standard (RS) and 0.05
57 mg/mL each of Raffinose RS, Glucose RS and Fructose RS in *Diluent*

58 *Impurity Standard solution:* 0.05mg/mL each of Sucrose RS, Raffinose RS, Glucose
59 RS and Fructose RS in *Diluent*

60 *Sample solution:* 10 mg/mL of Sucrose in *Diluent*

61 *Mobile phase:* Degassed water

62 Chromatographic system

63 • Mode: LC

64 • Detector: Refracted Index, maintained at a constant temperature (40°C)

65 • Column:

66 - Size: 7.8-mm \times 30-cm

67 - Strong cation-exchange resin consisting of sulfonated cross-linked

68 styrenedivinybenzene copolymer in the calcium form (9- μ m) *

69 [*Note: Aminex HPX-87C is suitable.]

70 - Temperature: 80 \pm 1°C

- 71 • Flow rate: 0.5 mL/min
- 72 • Injection volume: 10 μ L
- 73 • Run time: 30 min

74

75 System suitability requirements:

76 - Peak-to-valley ratio:

- 77 • NLT 2.5 between Raffinose and Sucrose

78 where H_p = height above the baseline of the raffinose, and H_v = height above
79 the baseline of the lowest point of the curve separating the impurity peak from
80 the sucrose peak in the *System suitability solution*.

81 - Resolution:

- 82 • NLT 1.5 between Sucrose and Glucose

83 [Notes:

- 84 1) The relative retention time with reference to sucrose for
85 raffinose/theandrose, glucose and fructose are 0.9, 1.2, and 1.6,
86 respectively. The retention time for Sucrose is about 10 min.
- 87 2) Raffinose is present in sucrose obtained from sugar beet and
88 Theandrose is present in sucrose obtained from sugar cane.]

89

- 90 - Relative standard deviation (%RSD): NMT 5.0%, for 6 replicate injections
91 of sucrose and each known impurity in the *Impurity Standard Solution*.

92

93 Injection: *Impurity Standard solution* and *Sample solution*

94 Calculate the percentage of each impurity in the portion of sucrose taken:

95

$$96 \quad \% \text{Impurity} = (r_u/r_s) \times (C_s/C_u) \times 100$$

97

98 r_u = peak response of each impurity from the *Sample solution*

99 r_s = peak response of each impurity from the *Impurity Standard solution*

100 C_s = concentration of each corresponding impurity RS in the *Impurity*
101 *Standard solution* (mg/mL);

102 C_u = concentration of the *Sample solution* (mg/mL)

103 [Note: Unknown impurities are calculated by comparing the unknown
104 impurity peak area to the sucrose peak area in the *Impurity Standard*

105 *solution.*]

106 Acceptance criteria:

- 107 - Individual known impurity (Raffinose/Theanderose, Glucose, Fructose):
- 108 NMT 0.2%
- 109 - Unknown impurities: NMT 0.10%
- 110 - Total impurities: NMT 1.0%
- 111 - Reporting threshold: 0.05%

112

113 **Dextrins.** If intended for use in the preparation of large-volume infusions, it complies
114 with the test for dextrins. To 2 ml of solution S add 8 ml of *water*, 0.05 ml of *dilute*
115 *hydrochloric acid* (73 g/l of HCl) and 0.05 ml of *0.05 M iodine*. The solution remains
116 yellow.

117

118 **Reducing sugars.** To 5 ml of solution S in a test-tube about 150 mm long and 16
119 mm in diameter add 5 ml of *water*, 1.0 ml of *1 M sodium hydroxide* and 1.0 ml of a 1
120 g/l solution of *methylene blue*. Mix and place in a water-bath. After exactly 2 min, take
121 the tube out of the bath and examine the solution immediately. The blue colour does
122 not disappear completely. Ignore any blue colour at the air/solution interface.

123

124 **Sulphite:** maximum 10 ppm calculated as SO₂.

125 Determine the sulphite content by a suitable enzymatic method based on the
126 following reactions. Sulphite is oxidised by sulphite oxidase to sulphate and hydrogen
127 peroxide which in turn is reduced by nicotinamide-adenine dinucleotide-peroxidase in
128 the presence of reduced nicotinamide-adenine dinucleotide (NADH). The amount of
129 NADH oxidised is proportional to the amount of sulphite.

130

131 *Test solution.* Dissolve 4.0 g of the substance to be examined in freshly prepared
132 *distilled water* and dilute to 10.0 ml with the same solvent.

133

134 *Reference solution.* Dissolve 4.0 g of the substance to be examined in freshly
135 prepared *distilled water*, add 0.5 ml of *sulphite standard solution (80 ppm SO₂)*
136 and dilute to 10.0 ml with freshly prepared *distilled water*. Use freshly prepared
137 *distilled water* as a blank.

138 Separately introduce 2.0 ml each of the test solution, the reference solution and
139 the blank in 10 mm cuvettes and add the reagents as described in the kit
140 instructions. Measure the absorbance at the maximum at about 340 nm before and
141 at the end of the reaction time and subtract the value obtained with the blank.

142 The absorbance difference of the test solution is not greater than half the
143 absorbance difference of the reference solution.

144

145 **Loss on drying** : maximum 0.1 per cent, determined on 2.000 g, by heating in an
146 oven at 105 °C for 3 h.

147

148 **Bacterial endotoxins** : less than 0.25 IU/mg, if intended for use in the
149 preparation of large-volume infusions.

150

151 **ASSAY**

152 Liquid chromatography as described in the test for related substances with the following
153 modifications.

154

155 *Standard solution*: 10 mg/mL of Sucrose RS in water

156 System suitability requirements:

- 157 - Peak-to-valley ratio and resolution: the same as those in the test for related
158 substances
- 159 - Relative standard deviation (%RSD): NMT 0.73%, for 5 replicate injections of
160 *Standard solution*.

161

162 Injection: *Standard solution* and *Sample solution*

163 Calculate the percentage of Sucrose (C₁₂H₂₂O₁₁) in the portion of sucrose taken:

164

$$165 \quad \% \text{Assay} = (r_U/r_S) \times (C_S/C_U) \times 100$$

166

167 r_U = peak response of Sucrose from the *Sample solution*

168 r_S = peak response of Sucrose from *Standard solution*

169 C_S = concentration of Sucrose RS in *Standard solution* (mg/mL)

170 C_U = concentration of the *Sample solution* (mg/mL)

171

172 Acceptance criteria: 98.0%–102.0%

173

174 **LABELLING**

175 The label states, where applicable, that the substance is suitable for use in the
176 manufacture of large-volume parenteral dosage forms.

177

178

179

180 **Reagents**

181
182
183
184
185
186
187
188
189
190
191
192
193
194
195
196
197
198
199
200
201
202
203
204
205
206
207
208
209

Hydrazine sulphate solution. Dissolve 1.0 g of hydrazine sulphate in water and dilute to 100.0 ml with the same solvent. Allow to stand for 4-6 h.

Hexamethylenetetramine solution. In a 100 ml ground-glass-stoppered flask, dissolve 2.5 g of hexamethylenetetramine in 25.0 ml of water.

Primary opalescent suspension (formazin suspension). To the hexamethylenetetramine solution in the flask add 25.0 ml of the hydrazine sulphate solution. Mix and allow to stand for 24 h. This suspension is stable for 2 months, provided it is stored in a glass container free from surface defects. The suspension must not adhere to the glass and must be well mixed before use.

Standard of opalescence. Dilute 15.0 ml of the primary opalescent suspension to 1000.0 ml with water. This suspension is freshly prepared and may be stored for up to 24 h.

Reference suspension I. To 5.0 ml of standard of opalescence add 95.0 ml of water. Mix and shake before use.

Sulphite standard solution (80 ppm SO₂).

Dissolve 3.150 g of *anhydrous sodium sulphite* in freshly prepared *distilled water* and dilute to 100.0 ml with the same solvent. Dilute 0.5 ml to 100.0 ml with freshly prepared *distilled water*.