# 1 2.28 Circular Dichroism Spectroscopy

The circular dichroism spectroscopy is a method used to analyze and determine the structure of optically active substances, discriminate active substances from enantiomers, diastereomers, etc by using the phenomenon (circular dichroism) in which the degrees of absorption of left and right circularly polarized lights differ in the absorption wave-

8 length region of active substances.

9 In this method circular dichroism is measured as the dif-10 ference of absorbance of left and right circularly polarized 11 lights as follows.

12 
$$\Delta A =$$

13 $\Delta A$ : Difference of absorbance of left and right circularly14polarized lights

 $A_{\rm L} - A_{\rm R}$ 

15  $A_{\rm L}$ : Absorbance of left circularly polarized light

16  $A_{\rm R}$ : Absorbance of right circularly polarized light

17 Also, the difference of molar absorption coefficients for

- 18 left and right circularly polarized lights can be expressed as
- 19 the molar circular dichroism as follows.

20 
$$\Delta \varepsilon = \varepsilon_{\rm L} - \varepsilon_{\rm R} = \frac{\Delta A}{c \times l}$$

21  $\Delta \varepsilon$ : Molar circular dichroism ((mol/L)<sup>-1</sup> • cm<sup>-1</sup>)

22  $\varepsilon_{L}$ : Molar absorption coefficient for left circularly polar-23 ized light ((mol/L)<sup>-1</sup> • cm<sup>-1</sup>)

24  $\varepsilon_{\rm R}$ : Molar absorption coefficient for right circularly po-25 larized light ((mol/L)<sup>-1</sup> · cm<sup>-1</sup>)

26 *c*: Concentration of an optically active substance in solu-27 tion (mol/L)

28 *l*: Path length (cm)

The following unit can also be used as the unit indicatingcircular dichroism.

 $g = \frac{\Delta \varepsilon}{\varepsilon}$ 

31 Dissymmetry factor (g factor):

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- 33  $\varepsilon$ : Molar absorption coefficient
- 34 Molar ellipticity:

35 In some apparatuses, circular dichroism is expressed in

36 units of ellipticity (°). In such a case, the molar ellipticity  $[\theta]$ 37 is calculated using the following equation.

$$[\theta] = \frac{\theta}{10 \times c \times l}$$

- 39 [ $\theta$ ]: Molar ellipticity ( $^{\circ} \cdot \text{cm}^2/\text{dmol}$ )
- 40  $\theta$ : Value (m°) of ellipticity calculated by apparatus
- 41 *c*: Concentration of an optically active substance in solu42 tion (mol/L)

43 *l*: Path length (cm)

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44 Molar ellipticity is related with molar circular dichroism45 by the following equation.

$$[\theta] = 2.303 \varDelta \varepsilon \ \frac{4500}{\pi} \approx 3300 \varDelta \varepsilon$$

47 Molar circular dichroism and molar ellipticity are often 48 used for analysis of peptides, proteins and nucleic acids. In 49 this case, mean residue weight, which is the molecular mass 50 divided by the number of monomeric residues, is used in the 51 calculation of molar concentration(c).

52 Mean residue weight

53 
$$=\frac{\text{molecular mass}}{\text{number of amino acid residues or nucleotide residues}}$$

54 Mean residue weight is 100 – 120 (generally 115) for 55 peptides and proteins, and is about 330 as sodium salt for 56 nucleic acids.

# 57 1. Apparatus

58 A circular dichroism spectrophotometer is used. A xenon 59 lamp is used as the light source. Light from the light source 60 is polarized at the time when being split by a double mono-61 chromator equipped with a crystal prism, resulting in mon-62 ochromatic linearly polarized light. The slit at the exit of the 63 monochromator eliminates extraordinary light. The mono-64 chromatic linearly polarized light is passed through a pho-65 toelastic modulator to be alternately modulated into left and 66 right circularly polarized lights at a constant frequency and is irradiated to a sample. 67

68 After the light that has passed through a sample to be 69 tested reaches a photomultiplier tube, the light is divided 70 into two electrical signals and amplified. One is the direct 71 current signal,  $V_{DC}$ , which reflects the light absorption of the 72 sample. The other is the alternating current signal,  $V_{AC}$ , 73 which occurs when the sample has circular dichroism and 74 has the same frequency as the modulation frequency of the 75 photoelastic modulator. The phase of the direct current signal indicates the sign of the circular dichroism (+ or -), and 76 77 the magnitude of the amplitude indicates the intensity of the 78 circular dichroism. Here,  $V_{AC}/V_{DC}$  is proportional to the dif-79 ference of the absorbances for left and right circular polar-80 ized lights,  $\Delta A$ . Generally, the wavelength range measured 81 by a circular dichroism spectrophotometer is about 170 to 82 800 nm, but some apparatuses can measure at a wider 83 wavelength range.

# 84 2. Methods

85 Set temperature, wavelength, path length and sample
86 concentration for measurement. Dissolve a sample in an
87 appropriate solvent, place it in a cell, and measure. In the
88 sample preparation, confirm the influence of impurities on
89 the spectrum, the structural change of the sample depending

90 on the concentration, the absorption of the solvent itself, 142 91 and the influence of the solvent on the sample structure. 143 92 Attention should be taken for the optical path length of a 144 93 sample cell, especially when the optical path length is short. 145 94 Furthermore, it should be noted for the absorption of light 146 by a sample because it may reduce a signal reaching a de-95 147 96 tector.

#### 97 2.1. Identification test

98 Specify molar circular dichroism or molar ellipticity 99 along with the wavelength at which it is maximum. The 100 identity of a substance can be confirmed when the molar 101 circular dichroism or molar ellipticity at the specified max-102 imum wavelength of the substance to be confirmed meets 103 this specification. Or, when the spectrum of a sample is 104 compared with the reference spectrum of the substance to be confirmed or the spectrum of the reference standard, and 105 106 both spectra give the same intensity of molar circular dichroism or molar ellipticity at the same wavelength, their 107 108 identity can be confirmed mutually.

## 2.2. Analysis of secondary structure 109

110 For peptides and proteins, specific spectra appear in the 111 far ultra-violet region. The secondary structure of peptides and proteins can be estimated by measuring the spectrum 112 below about 250 nm. Furthermore, it is possible to estimate 113 114 the three-dimensional structure from the near ultra-violet 115 spectrum. However, it should be noted that circular dichro-116 ism measurement observes the average property of a whole 117 molecule. For a  $\alpha$ -helix structure, negative maxima appear generally at 208 nm and 222 nm and a positive maximum 118 119 between 191 nm and 193 nm, for a  $\beta$ -sheet structure, a neg-120 ative maximum appears between 216 nm and 218 nm and a positive maximum between 195 nm and 200 nm, and for an 121 122 irregular structure, a negative maximum appears between 123 195 nm and 200 nm. Methods for analyzing the proportion 124 of secondary structures from a circular dichroism spectrum include a method using a calculation formula and a method 125 using a database. It can also be calculated by multivariate 126 127 analysis. Whenever any method is used, the method used 128 for the calculation is specified in the test method.

#### 129 3. Verification of the performance of apparatus

130 A wavelength-calibrated apparatus is used, and the per-131 formance of the apparatus is verified using a sample with 132 quality suitable for the measurement of circular dichroism 133 and with known  $\Delta \epsilon$ .

## 3.1. Accuracy of circular dichroism 134

135 Calibrate the accuracy of circular dichroism with a sub-136 stance with known  $\Delta \varepsilon$ , such as isoandrosterone, ammonium 137 d-camphorsulfonate, etc. (substances recommended by the apparatus manufacturer may be used). When using iso-138 139 androsterone, weigh exactly 10.0 mg of isoandrosterone, 140 and dissolve in ethanol (99.5) to make exactly 10 mL. When 141 the circular dichroism spectrum of the prepared solution is

measured in the range of 280 nm to 360 nm using a cell with a path length of 10 mm,  $\Delta \varepsilon$  at 304 nm is +3.3.

## 3.2. Linearity of modulation

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Calibrate the linearity of modulation with a substance with known  $\Delta \varepsilon$ , such as ammonium *d*-camphorsulfonate (substances recommended by the apparatus manufacturer 148 may be used). When using ammonium *d*-camphorsulfonate, 149 weigh exactly 6.0 mg of ammonium d-camphorsulfonate 150 and dissolve in water to make exactly 10 mL. When the circular dichroism spectrum of the prepared solution is 152 measured in the range of 185 nm to 340 nm using a cell 153 with a path length of 1 mm,  $\Delta \varepsilon$  at 290.5 nm is +2.2 to +2.5 154 and  $\Delta \varepsilon$  at 192.5 nm is -4.3 to -5.