## **Esomeprazole Magnesium Hydrate** 1

エソメプラゾールマグネシウム水和物 2



4 C34H36MgN6O6S2.3H2O: 767.17

- Monomagnesium bis[(S)-5-methoxy-2-{[(4-methoxy-5
- 6 3,5-dimethylpyridin-2-yl)methyl]sulfinyl}-1H-
- 7 benzimidazol-1-ide] trihydrate

8 [217087-09-7]

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Esomeprazole Magnesium Hydrate contains not less 10 than 98.0% and not more than 102.0% of esomeprazole 11

magnesium (C<sub>34</sub>H<sub>36</sub>M<sub>g</sub>N<sub>6</sub>O<sub>6</sub>S<sub>2</sub>: 713.12), calculated on 12

the anhydrous basis. 13

14 **Description** Esomeprazole Magnesium Hydrate occurs as 15 a white or slightly colored powder.

16 It is sparingly soluble in methanol, and slightly soluble in 17 ethanol (99.5) and in water.

Optical rotation <2.49>  $[\alpha]_D^{20}$ : +137 - +142° (0.25 g, 18 19 methanol, 25 mL, 100 mm).

**Identification** (1) Determine the absorption spectrum of 20 a solution of Esomeprazole Magnesium Hydrate in methanol 21 22 (1 in 50000) as directed under Ultraviolet-visible Spectro-23 photometry <2.24>, and compare the spectrum with the Reference Spectrum: both spectra exhibit similar intensities of 24 25 absorption at the same wavelengths.

26 Determine the infrared absorption spectrum of (2)27 Esomeprazole Magnesium Hydrate as directed in the potas-28 sium bromide disk method under Infrared Spectrophotometry 29 <2.25>, and compare the spectrum with the Reference Spec-30 trum: both spectra exhibit similar intensities of absorption at

31 the same wave numbers. 32 (3) Ignite 0.5 g of Esomeprazole Magnesium Hydrate as

33 directed under Residue on Ignition <2.44>, cool, and dissolve the residue in 10 mL of water: the solution responds to Qual-34 itative Tests <1.09> (1) for magnesium salt. 35

Purity (1) Clarity and color of solution (color)-Con-36 37 duct this procedure within 1 hour after preparation of the 38 sample solution. Dissolve 0.20 g of Esomeprazole Magne-39 sium Hydrate in 10 mL of methanol, filter this solution 40 through a membrane filter with a pore size not exceeding 0.5  $\mu$ m, and use the filtrate as the sample solution. Determine the 41 42 absorbance of the sample solution at 440 nm as directed un-43 der Ultraviolet-visible Spectrophotometry <2.24>: the ab-

44 sorbance is not more than 0.2. 45 Clarity and color of solution (Turbidity)-Being (2)46 specified separately when the drug is granted approval based 47 on the Law.

48 (3) Related substances—Conduct this procedure within 49 10 minutes after preparation of the sample solution. To 5 mg of Esomeprazole Magnesium Hydrate add 20 mL of the mo-50 51 bile phase, sonicate to dissolve in a water bath with occa-52 sional shaking, add the mobile phase to make 25 mL and use 53 this solution as the sample solution. Perform the test with 50 54  $\mu$ L of the sample solution as directed under Liquid Chroma-55 tography <2.01> according to the following conditions. De-56 termine each peak area by the automatic integration method, 57 and calculate their amounts by the area percentage method: 58 the amount of the peak of related substance D having the rel-59 ative retention time of about 0.8 to esomeprazole is not more 60 than 0.2%, the amount of the peak of related substance E hav-61 ing the relative retention time of about 0.4 is not more than 0.1%, and the amount of the peak other than esomeprazole 62 63 and the peaks mentioned above is not more than 0.10%. Fur-64 thermore, the total amount of the peaks other than esomepra-65 zole is not more than 0.5%.

66 **Operating** conditions—

67 Detector: An ultraviolet absorption photometer (wave-68 length: 280 nm).

Column: A stainless steel column 3.9 mm in inside diam-69 70 eter and 15 cm in length, packed with octadecylsilanized sil-71 ica gel for liquid chromatography (4  $\mu$ m in particle diameter).

72 Column temperature: A constant temperature of about 73 22°C.

74 Mobile phase: To 5.2 mL of a solution of sodium dihydro-75 gen phosphate dihydrate (39 in 250) and 63 mL of 0.5 mol/L 76 disodium hydrogen phosphate TS add water to make 1000 77 mL. To 250 mL of this solution add water to make 1000 mL. 78 To 760 mL of this solution add 240 mL of acetonitrile.

79 Flow rate: Adjust so that the retention time of esomepra-80 zole is about 9 minutes.

Time span of measurement: About 4.5 times as long as the 82 retention time of esomeprazole.

## 83 System suitability-

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84 Test for required detectability: Dissolve 1 mg each of 85 Omeprazole RS and omeprazole sulfone in the mobile phase 86 to make 25 mL, and use this solution as the solution for sys-87 tem suitability test. Pipet 2.5 mL of the solution for system 88 suitability test, and add the mobile phase to make exactly 100 89 mL. Pipet 1 mL of this solution, add the mobile phase to 90 make exactly 10 mL. When the procedure is run with 50  $\mu$ L 91 of this solution under the above operating conditions, the SN 92 ratio of the peak of omeprazole is not less than 10.

93 System performance: When the procedure is run with 50 94  $\mu$ L of the solution for system suitability test under the above 95 operating conditions, omeprazole sulfone and omeprazole 96 are eluted in this order with the resolution between these 147 97 peaks being not less than 3.

98 Enantiomer Conduct this procedure within 1 hour after preparation of the sample solution. Dissolve 40 mg of 99 100 Esomeprazole Magnesium Hydrate in 5 mL of methanol, and add the diluting solution to make 25 mL. To 0.2 mL of this 101 solution add the diluting solution to make 10 mL, and use this 102 103 solution as the sample solution. Perform the test with 20  $\mu$ L 104 of the sample solution as directed under Liquid Chromatog-105 raphy <2.01> according to the following conditions. Deter-106 mine each peak area by the automatic integration method, 107 and calculate their amounts by the area percentage method: 108 the amount of the peak of related substance F (enantiomer) 109 having the relative retention time of about 0.7 to esomeprazole is not more than 0.2%. 110 111 Diluting solution: To 11 mL of a solution of trisodium 112 phosphate dodecahydrate (19 in 200) and 22 mL of 0.5 mol/L 113 disodium hydrogen phosphate TS add water to make 1000 114 mL. 115 Operating conditions— 116 Detector: An ultraviolet absorption photometer (wavelength: 302 nm). 117 Column: A stainless steel column 4.0 mm in inside diam-118 eter and 10 cm in length, packed with  $\alpha_1$ -acid glycoprotein 119 binding silica gel for liquid chromatography (5  $\mu$ m in particle 120 121 diameter). 172 122 Column temperature: A constant temperature of about 123 22°C. 124 Mobile phase: To 70 mL of a solution of sodium dihydrogen phosphate dihydrate (39 in 250) and 20 mL of 0.5 mol/L 125 126 disodium hydrogen phosphate TS add water to make 1000 127 mL. To 250 mL of this solution add water to make 1000 mL. 128 To 850 mL of this solution add 150 mL of acetonitrile. 129 Flow rate: Adjust so that the retention time of esomepra-130 zole is about 5 minutes. 131 System suitability-132 System performance: Dissolve 2 mg of Omeprazole RS in 133 the diluting solution to make 10 mL. To 0.2 mL of this solution add the diluting solution to make 10 mL. When the pro-134 135 cedure is run with 20  $\mu$ L of this solution under the above operating conditions, the resolution between the peaks of re-136

137 lated substance F (enantiomer) having the relative retention

138 time of about 0.7 to esomeprazole and esomeprazole is not 139 less than 3.

140 Water  $\langle 2.48 \rangle$  6.0 – 8.0% (0.2 g, volumetric titration, direct 141 titration).

142 Magnesium Weigh accurately about 0.4 g of Esomepra-

zole Magnesium Hydrate, add 25 mL of methanol, and soni-143

193 144 cate to dissolve. To this solution add 25 mL of water, 10 mL

145 of ammonia solution (28) and 20 mL of 0.05 mol/L disodium 194

dihydrogen ethylenediamine tetraacetate VS, and titrate 146

<2.50> the excess disodium dihydrogen ethylenediamine tetraacetate with 0.05 mol/L zinc sulfate VS (indicator: 50 mg of eriochrome black T-sodium chloride indicator) until 150 the color of the solution changes from blue to purple. Perform a blank determination in the same manner: Esomeprazole Magnesium Hydrate contains 3.30 - 3.55% of magnesium 152 (Mg: 24.31) calculated on the anhydrous basis.

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Each mL of 0.05 mol/L disodium dihydrogen ethylenediamine tetraacetate VS = 1.215 mg of Mg

Assay Weigh accurately about 10 mg of Esomeprazole Magnesium Hydrate, dissolve in 10 mL of methanol, add 10 mL of the diluting solution, add water to make exactly 200 mL, and use this solution as the sample solution. Separately, weigh accurately about 10 mg of Omeprazole RS (separately determine the loss on drying <2.41> in vacuum at 50°C for 2 hours using phosphorous (V) oxide as a desiccant, using 1 g of Omeprazole RS), dissolve in 10 mL of methanol, add 10 mL of the diluting solution, add water to make exactly 200 mL, and use this solution as the standard solution. Perform the test with 20  $\mu$ L each of the sample solution and standard solution as directed under Liquid Chromatography <2.01> according to the following conditions, and determine the peak area,  $A_{\rm T}$ , of esomeprazole obtained from the sample solution and the peak area,  $A_{\rm S}$ , of omeprazole from the standard solution.

Amount (mg) of esomeprazole magnesium  $(C_{34}H_{36}M_gN_6O_6S_2)$  $=M_{\rm S} \times A_{\rm T}/A_{\rm S} \times 713.12/(345.42 \times 2)$ 

713.12: Molecular mass of esomeprazole magnesium 345.42: Molecular mass of omeprazole

 $M_{\rm S}$ : Amount (mg) of Omeprazole RS taken, calculated on the dried basis

Diluting solution: To 110 mL of a solution of trisodium phosphate dodecahydrate (19 in 200) and 220 mL of 0.5 mol/L disodium hydrogen phosphate TS add water to make 1000 mL.

Operating conditions—

Detector: An ultraviolet absorption photometer (wavelength: 280 nm).

Column: A stainless steel column 4.0 mm in inside diameter and 125 mm in length, packed with octylsilanized silica gel for liquid chromatography (5  $\mu$ m in particle diameter).

Column temperature: A constant temperature of about 22°C.

Mobile phase: To 5.2 mL of a solution of sodium dihydrogen phosphate dihydrate (39 in 250) and 63 mL of 0.5 mol/L disodium hydrogen phosphate TS add water to make 1000

mL. To 163 mL of this solution and 487 mL of water add 350 227
mL of acetonitrile. 228

- Flow rate: Adjust so that the retention time of esomeprazole is about 4 minutes.
- 199 System suitability—
- 200 System performance: When the procedure is run with 20
- 201  $\mu$ L of the standard solution under the above operating condi-
- 202 tions, the number of theoretical plates and the symmetry fac-
- 203 tor of the peak of omeprazole are not less than 2000 and not
- 204 more than 1.5, respectively.
- 205 System repeatability: When the test is repeated 6 times 206 with 20  $\mu$ L of the standard solution under the above operating
- with 20  $\mu$ L of the standard solution under the above operating conditions, the relative standard deviation of the peak area of
- 208 omeprazole is not more than 1.0%.

209 Containers and storage Containers—Tight containers.

- 210 Storage—Light-resistant.
- 211 Others
- 212 Related substances D (omeprazole sulfone):
- 213 5-Methoxy-2-{[(4-methoxy-3,5-dimethylpyridin-2-yl)me-
- 214 thyl]sulfonyl}-1H-benzimidazole



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- 216 Related substances E:
- 217 4-Methoxy-2-{[(S)-(5-methoxy-1H-benzimidazol-2-
- 218 yl)sulfinyl]methyl}-3,5-dimethylpyridine 1-oxide



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- 220 Related substances F (enantiomer):
- 221 (R)-5-Methoxy-2-{[(4-methoxy-3,5-dimethylpyridin-2-
- 222 yl)methyl]sulfinyl}-1*H*-benzimidazole



224 Add the following to 9.01 Reference 225 Standards (1).

226 Omeprazole RS

## 227 Add the following to 9.41 Reagents, Test 228 Solutions.

## 229 **Omeprazole sulfone** C<sub>17</sub>H<sub>19</sub>N<sub>3</sub>O<sub>4</sub>S : 361.42

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*Description* It occurs as a white to brown powder.

231 Identification Determine the 1H spectrum of a solution of 232 Omeprazole sulfone in deuterated dimethyl sulfoxide for nu-233 clear magnetic resonance spectroscopy (1 in 100) as directed 234 under Nuclear Magnetic Resonance Spectroscopy <2.21>, us-235 ing tetramethylsilane for nuclear magnetic resonance spec-236 troscopy as an internal reference compound: it exhibits a sin-237 glet signal A at around  $\delta$  2.17 ppm, a singlet signal B at 238 around  $\delta$  2.20 ppm, a singlet signal C at around  $\delta$  3.68 ppm, 239 a singlet signal D at around  $\delta$  3.82 ppm, a singlet signal E at 240 around  $\delta$  5.01 ppm, a broad singlet signal F at around  $\delta$  7.61 241 ppm, and a singlet signal G at around  $\delta$  8.04 ppm. The ratio 242 of the integrated intensity of each signal, A:B:C:D:E:F:G is

about 3:3:3:3:2:1:1 (When the frequency is 500 MHz).